

Formulation of pH-responsive lipid-polymer hybrid nanoparticles for co-delivery and enhanced antibacterial activity of 18 β -glycyrrhetic acid and vancomycin against MRSA

By

Yajna Jaglal

(BSc Medical Science (Hon.) – University of KwaZulu-Natal)

Student number: 216000829

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Supervisor: Professor Thirumala Govender

(Ph.D., University of Nottingham, Nottingham, United Kingdom)

Co-supervisor: Dr Calvin Omolo

(Ph.D., University of Kwazulu-Natal, Durban, South Africa)

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“The only way to do great work is to love what you do.”

- Steve Jobs-

"This dissertation is dedicated to my grandparents, parents, brother, mentors and best friends for their undivided love, support and great advice.

I appreciate you all."

Declaration 1 – Plagiarism

I, Miss Yajna Jaglal declare that

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Declaration 2 – Publications

Details of contribution to publications that form part and/or include research presented in this dissertation:

**Yajna Jaglal, Nawras Osman, Calvin A. Omolo, Chunderika Mocktar, Nikita Devnarain, Thirumala Govender. Formulation of pH-responsive lipid-polymer hybrid nanoparticles for co-delivery and enhancement of antibacterial activity of vancomycin and 18 β -glycyrrhetic acid. SUBMITTED MANUSCRIPT.
Reference number: IJP-S-20-04690**

Miss Yajna Jaglal contributed to the project design and characterization of the novel nanoformulation. In addition, Miss Jaglal was responsible for the analysis and interpretation of all data, wrote the first draft of the paper and undertook all corrections. Mrs Nawras Osman assisted with formula optimization and characterization and *in vitro* drug release studies. Dr Calvin Omolo contributed to conceptualisation, design and formulation of the drug delivery system and supervision of the characterization and application studies and thesis and abstract editing. Dr Chunderika Mocktar supervised the *in vitro* antibacterial activity studies. Dr Nikita Devnarain conducted the cytotoxicity studies, *in silico* studies and editing. Prof Thirumala Govender served as supervisor and was responsible for project conceptualization, thesis and abstract editing and overall supervision of the study.

Research output from the dissertation

Submitted Manuscript

The following research paper was submitted to the International Journal of Pharmaceutics which has an impact factor of 4.845 from work done during this study.

Yajna Jaglal, Nawras Osman, Calvin A. Omolo, Chunderika Mocktar, Nikita Devnarain, Thirumala Govender. Formulation of pH-responsive lipid-polymer hybrid nanoparticles for co-delivery and enhancement of antibacterial activity of vancomycin and 18 β -glycyrrhetic acid. SUBMITTED MANUSCRIPT. Reference number: IJP-S-20-04690

* The submitted manuscript can be found in Chapter 3.

Abstract

Background: Due to the rise in antimicrobial resistance and the challenges accompanied by conventional antibiotic dosage forms, there is a need for developing drug delivery systems that enhance, protect and potentiate the current antibiotics in the market. Furthermore, natural derivatives from plants have proven to be potent antimicrobial agents. Therefore, their combination with antibiotics could be effective in overcoming antimicrobial resistance.

Aim: The aim of this study was to co-deliver vancomycin and 18 β -glycyrrhetic acid via pH-responsive lipid-polymer hybrid nanoparticles (VCM-GAPAH-LPHNPs) formulated from polyallylamine and oleic acid (OA) and to explore its potential for enhanced activity and targeted delivery.

Methods: Molecular dynamics and stability studies were used to determine the stability of the oil and water phases independently as well as VCM-GAPAH-LPHNPs as a complex. VCM-GAPAH-LPHNPs were prepared using the micro-emulsion technique. The size, polydispersity index and zeta potential of VCM-GAPAH-LPHNPs were determined using the dynamic light scattering technique. Transmission electron microscopy analysis was conducted to determine the morphology of VCM-GAPAH-LPHNPs. The entrapment efficiency and drug loading were determined using the ultrafiltration method. Differential scanning calorimetry was used to determine the thermal profiles of VCM-GAPAH-LPHNPs and its components. *In vitro* drug release studies were performed using the dialysis bag technique. Drug release kinetics were analysed using the DDSolver program. Cytotoxicity of VCM-GAPAH-LPHNPs were determined using the MTT assay. Haemolysis of VCM-GAPAH-LPHNPs were performed at different concentrations using sheep blood. *In vitro* antibacterial activity of VCM-GAPAH-LPHNPs were determined against SA and methicillin-resistant *Staphylococcus aureus* (MRSA) at pH 6 and 7.4. Time killing assay was performed using the plate colony count method. MRSA biofilm study was performed using the crystal violet assay.

Results: Molecular dynamics indicated VCM-GAPAH-LPHNPs to be stable. VCM-GAPAH-LPHNPs were successfully prepared using the micro-emulsion technique. VCM-GAPAH-LPHNPs size, polydispersity index, zeta potential and encapsulation

efficiency were found to be 198.4 ± 0.302 nm, 0.255 ± 0.003 , -3.8 ± 0.335 mV and 69.46 ± 2.52 % respectively. Thermal profiles of lyophilized VCM-GAPAH-LPHNPs showed transformation from crystallization to amorphous form. *In vitro* drug release studies revealed that VCM-GAPAH-LPHNPs released 60% of VCM after 24 h whereas bare VCM released 90% of VCM after 24 h hence VCM-GAPAH-LPHNPs showed sustained drug release compared to bare VCM. At pH 6 VCM-GAPAH-LPHNPs released 82% of VCM after 24 h whereas at pH 7.4 VCM-GAPAH-LPHNPs released 60% of VCM after 24 h indicating VCM-GAPAH-LPHNPs had a faster drug release at pH 6 compared to pH 7.4. The Weibull model was considered the best fit model for VCM-GAPAH-LPHNPs. The MTT assay revealed 75% > cell viability which indicated VCM-GAPAH-LPHNPs to be non-cytotoxic. At 0.5 mg/ml VCM-GAPAH-LPHNPs showed < 1% haemolysis. Stability studies at 4 °C and room temperature indicated VCM-GAPAH-LPHNPs to be stable. *In vitro* antibacterial activity against MRSA treated with VCM-GAPAH-LPHNPs demonstrated a 16-fold lower minimum inhibitory concentration than bare VCM at acidic conditions. The time-killing assay study at 12 h revealed that VCM-GAPAH-LPHNPs eliminated 100% of MRSA cells whereas bare VCM eliminated 55% of MRSA cells. The crystal violet assay analysis revealed VCM-GAPAH-LPHNPs ability to eliminate MRSA biofilms.

Conclusion: VCM-GAPAH-LPHNPs could effectively treat MRSA infections at a faster rate as compared to bare VCM. Therefore, this novel pH-responsive LPHNPs may serve as a promising nanocarrier for enhancing antibiotic delivery and antibacterial activity.

Keywords: pH-responsive; lipid-polymer hybrid nanoparticles; vancomycin; methicillin-resistant *Staphylococcus aureus*; 18 β -glycyrrhetic acid

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List of acronyms

Ala	Alanine	PAH	Polyallylamine hydrochloride
CACO-2	Human intestinal epithelial cancer cells	PDI	Polydispersity index
CFU	Colony forming units	PGA	Polyglycolides
CO ₂	Carbon dioxide	PLA	Poly lactides
CV	Crystal violet	PLGA	Poly lactide co-glycolides
DL	Drug loading	PNPs	Polymeric nanoparticles
DLS	Dynamic light scattering	PBS	Phosphate-buffered saline
DSC	Differential scanning calorimetry	RBCs	Red blood cells
EE	Encapsulation efficiency	RMSD	Root-mean-square deviation
GA	Glycyrrhetic acid	RMSE	Root-mean-square error
HepG2	Human liver adenocarcinoma cells	RMSF	Root-mean-square-fluctuation
HPLC	High-performance liquid chromatography	SA	<i>Staphylococcus aureus</i>
LPHNPs	Lipid-polymer hybrid nanoparticles	SEM	Scanning electron microscopy
MD	Molecular dynamics	SLNs	Solid lipid nanoparticles
MDR	Multidrug resistance	TB	Tuberculosis
MHA	Mueller Hinton Agar	TEM	Transmission electron microscopy
MHB	Mueller Hinton broth	TFA	Trifluoroacetic acid
MIC	Minimum inhibitory concentration	UA	Uranyl acetate
MRSA	Methicillin-resistant <i>Staphylococcus aureus</i>	UV	Ultraviolet
MTT	3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide	VCM	Vancomycin
NDDS	Nano drug delivery systems	XDR	Extreme drug resistance
OA	Oleic acid	ZP	Zeta potential

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Chapter One

Introduction

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Chapter One - Introduction

1.1 Introduction

This chapter provides a brief background of the study including the crisis of infectious diseases, limitations of conventional antibiotic therapy and the threat of a post-antibiotic era. It further discusses the advantages of pH-responsive nanodrug delivery systems and their role in combatting antibiotic resistance. This is followed by the aims, novelty and significance of the study and concludes with an overview of the dissertation.

1.2 Background

Infectious diseases, particularly bacterial infections are considered one of the primary causes of morbidity and mortality worldwide, despite influential research advancements (1). Recent data indicates that by 2050 there will be 10 million annual deaths caused by bacterial infections (2). Antibiotics are designed to inhibit or stop bacterial infections and have revolutionized the treatment of infectious diseases via their bacteriostatic and bactericidal effects (3). However, conventional antibiotic dosage forms need to be administered regularly for a sustained period to maintain adequate concentrations at target sites of infection (4) and to avoid persistent infections due to the development of antibiotic resistance (5). Furthermore, the high dose of antibiotics being administered leads to harmful side effects (6, 7), poor pharmacokinetic properties (6), a burden on the healthcare sector and treatment costs and poor patient compliance (8).

Such disadvantages, compounded by incorrect usage and exploitation of antibiotics (1, 9) have led to an antibiotic-resistant era (9). Antibiotic-resistant and multidrug-resistant (MDR) bacteria such as MRSA pose a greater threat to humankind than ever imagined (10). MRSA is a common pathogen present in community and hospital-acquired infections (11) and contributes to the development of sepsis (12, 13), peritonitis (13), endocarditis and bacteraemia (14). The prolonged discovery and advancement of novel antibiotics, as well as high production costs, emphasize the need to strategically introduce new dimensions to suppress the rapid increase of bacterial resistance to antibiotics and limitations of conventional antibiotics (15, 16). Nanoengineered antibiotic delivery systems have been reported as a promising approach to overcome the restrictions of conventional antibiotics and resistance of bacteria (17, 18). Nano drug delivery systems (NDDS) are defined as biocompatible, nanosized materials having a large surface area to mass ratio (19). There are several advantages that NDDS have over

conventional dosage forms of antibiotics, these include targeted antibiotic delivery to infection sites, enhanced localisation of the antibiotic within target tissues (20), improved pharmaceutical stability (21), improved antibiotic solubility, improved cellular absorption, sustained antibiotic release (22), improved patient compliance and reduced side effects (23).

Some nanosystems that have been explored as NDDS for antibiotics include liposomes (24), polymeric nanoparticles (PNPs) (25), lipid polymer hybrid nanoparticles (LPHNPs) (26), dendrimers (27), solid lipid nanoparticles (SLNs) (28) and micelles (29). Of these NDDS, lipid-polymer hybrid nanoparticles (LPHNPs) have shown major potential as antibiotic nanocarriers with enhanced antibacterial efficacy. They were first proposed in the late 1970s as a promising antibiotic delivery system (30) and are composed of biocompatible/biodegradable polymers, whereby the antibiotic is solubilized, encased or anchored to the outer surface of nanoparticles (31). There are many advantages of LPHNPs, which include improved concentration of antibiotics at target infection sites thus improving drug safety (32), protection of the antibiotic (23), reduced premature drug release before arrival at the infection site, controlled and sustained antibiotic release, improved cell penetration and solubility (33), cost-effective upscaled production using controlled polymerization techniques (34), improved stability of volatile chemical substances (35), antimicrobial properties (36) and biocompatibility with tissues (31-34, 36). Recently LPHNPs have been employed to deliver the messenger RNA from severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2) for creation of the vaccine against the disease (37). Therefore, LPHNPs are efficient and effective drug delivery nanosystems (38).

Spurred by the advancements in pharmaceutical chemistry and drug delivery, stimuli-responsive NDDS are of recent interest. Researchers are committed to developing stimuli-responsive nanosystems (39) to heighten the efficiency and effectiveness of NDDS. Stimuli-responsive antibiotic delivery systems are designed for optimal response to endogenous stimuli (i.e., enzymes, redox potential, ionic microenvironment and pH levels) and exogenous stimuli (i.e., temperature, ultrasound, electric, mechanical, light and magnetic fields) at an infection site, allowing targeted antibiotic release, improved antibiotic accumulation as well as enhanced bioavailability at the target site (39, 40). Among these stimuli, pH-responsive NDDS have become extremely popular in the literature. The significance of pH-responsive NDDS lies in their ability to deliver antibiotics when minor pH changes according to the pathophysiology properties of the disease are presented, resulting in enhanced therapeutic efficacy and patient compliance (40, 41). Cancer and infectious diseases due to bacterial infection, like tuberculosis,

present with changes towards an acidic pH, which varies from the physiological pH of 7.4 (41, 42). Effective pH-responsive nanosystems such as pH-responsive liposomes (24), polymeric nanoparticles (30) and micelles (29) aim at targeting acidic pH conditions (42). pH-sensitive NDDS have been reported for the delivery of anti-cancer (41-43) and antibacterial (44-46) agents at targeted infection sites. However, limited research has been done on pH-sensitive nano drug delivery to bacterial infection sites, specifically that of MRSA, which has the potential to massively influence infectious disease treatment by improved and targeted drug delivery to acidic infection sites and curb the limitations of conventional antibiotics (47, 48).

Currently, LPHNPs are being employed to efficiently deliver drugs, making them promising nanosystems (49). LPHNPs overcome the above-mentioned problems of conventional antibiotic dosage forms and antimicrobial resistance by combining the structural components of PNPs and liposomes (50). LPHNPs are core-shell self-assembled drug delivery systems fabricated from a hydrophobic lipid core and a polymeric matrix (51). The biofunctional properties of LPHNPs allow the formulation of a system that is stimuli-responsive (52). This is attained by focusing on the properties of the nanoparticle matrix in order to establish targeted drug delivery and enhanced antibacterial activity (53). The development in the field of nanotherapeutics and material sciences has stimulated the advancement of pH-sensitive nanosystems for effective and efficient antibiotic delivery (54, 55). Several pH-responsive LPHNPs have been reported for antibacterial and anticancer studies (56-58). However, there is limited literature on employing pH-sensitive LPHNPs for antibiotic delivery. Therefore, using this approach, systems can be fabricated to have programmable destabilization and drug release due to pH changes that correspond with bacterial infection sites (58, 59).

The bioactive phytochemical composites found in plants and natural plant derivatives possess a broad spectrum of activity, 18 β -glycyrrhetic acid (GA) is such a natural derivative (60). GA is the hydrolyzed product of glycyrrhizic acid which is derived from the *Glycyrrhiza glabra* (liquorice) plant (61). This biologically active compound has been reported to have anti-allergic (62), antibacterial, antiviral, antitumor and anti-inflammatory characteristics (63, 64). Due to the antibacterial activity of GA, it can be explored by co-delivery with antibacterial drugs in the market for synergistic action (63, 65).

Literature reports indicate the combinational therapy of antibiotics in the market and natural plant derivatives that have shown to improve antibacterial activity (66). However, there are no studies of GA being co-delivered with antimicrobial drugs in a NDDS to target bacterial

infections. Therefore, the combined delivery of GA and antibiotics can further be explored by co-loading in a stimuli-responsive nanosystem for targeted and enhanced antimicrobial activity (67). Such a system will also provide scientific advancement to the field of pharmaceuticals.

There are numerous reports of GA being co-delivered with cancer and inflammatory drugs in nanosystems for the treatment of cancer diseases (68-70). However, from our search of the literature, such a formulation for co-delivery of GA and antibiotics has not been done before. Thus, in this study, we propose the co-delivery of GA and VCM via polyallylamine hydrochloride (PAH) LPHNPs for enhanced and targeted antibiotic delivery. The pH-responsive LPHNPs matrix consists of PAH and oleic acid (OA). We envisage the pH-response will stem from the ionisable carboxylic acid groups of OA and primary amino groups of PAH. At a basic pH, carboxylic acid groups deprotonate and form electrostatic bonds with the amine groups of PAH. Thus, the system will have an overall negative charge. Consequently, at acidic conditions, both the amine and carboxylic groups will protonate, the carboxylic group will have a slightly positive charge that will repel the highly positive charge of PAH amine groups thus LPHNPs charge shift from negative to positive charge, followed by cleavage of the electrostatic bond, ultimately resulting in swelling of LPHNPs and faster release of drugs. Additionally, the positive charge of PAH polymer may promote attachment to the negatively charged bacterial cell wall, thus enhancing antibacterial activity. Such a system to the best of our knowledge has not been reported before. Moreover, LPHNPs can encapsulate both hydrophilic and hydrophobic drugs, in this system GA is co-delivered with VCM. Such a strategy has not been reported before for delivery of antibiotics with bioactive plant by-products. Therefore, this study will report novel multifunctional LPHNPs that are pH- responsive for co-delivery of GA and VCM.

1.3 Problem statement

Globally, bacterial infections are a significant problem in the healthcare sector. This is due to the several disadvantages of current treatment regimes, including conventional dosage forms of current antibiotics, low drug concentrations at infection sites, frequent high dosages, poor pharmacokinetics, poor patient compliance and adverse side effects. Such restrictions have resulted in the birth of an antibiotic-resistant era that is associated with a rise in mortality and morbidity rates. The advancement of new antimicrobials is deteriorating, urging the design and discovery of novel approaches to improve current antibiotic treatment. The establishment of effective pH-responsive antibiotic co-delivery nanosystems can obtain targeted delivery at

infection sites, thereby enhancing antibiotic therapy. Ultimately, the development of pH-responsive co-delivery systems is required to overcome the challenges associated with current antibiotic dosage forms and antimicrobial resistance.

1.4 Hypothesis

We hypothesize that the co-delivery of GA and VCM via pH-responsive PAH-LPHNPs formulation can enhance its antibacterial activity.

1.5 Aims and objectives

The aim of this study was to identify the potential of the co-delivery of GA and VCM via pH-responsive PAH lipid-polymer hybrid nanoparticles (GAPAH-LPHNPs) for enhancing antibacterial activity.

The objectives of the study were:

1. To synthesize novel pH-responsive GAPAH lipid-polymer hybrid nanoformulation encapsulating VCM and GA.
2. To optimize and characterize VCM-GAPAH-LPHNPs in terms of particle size, polydispersity index (PDI), zeta potential (ZP), pH-responsiveness, morphology, entrapment efficiency and *in vitro* drug release.
3. To assess *in vitro* antibacterial activity of VCM-GAPAH-LPHNPs against SA and MRSA.
4. To identify time-killing kinetics against MRSA.
5. To perform crystal violet assay analysis against MRSA biofilms.

1.6 Novelty of the study

The research conducted in this study is novel for the following reasons:

- This study reports for the first time LPHNPs coined from PAH and OA.
- This study reports the synthesis and characterization of multifunctional LPHNPs designed for pH-responsive co-delivery of VCM and GA, which have not been previously reported in the literature.
- This work also reports for the first time the co-delivery of an antibiotic and bioactive metabolic compound that enhanced the activity of the antibiotic in the market.

1.7 Significance of the study

These reported pH-responsive LPHNPs for co-delivery of GA and VCM present a novel and promising avenue for targeting acidic infection sites, thus enhancing drug localisation within target tissues, reducing concentration of the dosage needed for ideal treatment, improving antibiotic properties and preventing the occurrence of bacterial resistance. Ultimately, this results in minimal side effects and improves patient adherence to treatment. The potential significance of this study is mentioned below:

Advanced nanomedicine

This study proposes the novel pH-responsive LPHNPs for co-delivery of an antibiotic and natural compound formulation as a medicine. This nanomaterial and medication can play a role in pharmaceutical companies in the development of novel pH-responsive drug delivery vehicles that could be more effective than conventional antibiotics.

Advanced patient and disease treatment

This novel nanocarrier has the potential to advance bacterial infection treatment by allowing targeted and controlled release at infection sites, enhancing drug localization and bioavailability at acidic infection site thus, contributing to the improvement of antibacterial properties, reduction of dosage frequency and minimal side effects, which ultimately improves patient adherence and defeats threats against antibacterial resistance.

Creating novel scientific knowledge

This study can identify new scientific advancements in the preparation and characterization of the combinational delivery of antibiotics and natural plant compounds of pH-responsive lipid-polymer hybrid nanoparticles and their potential in the world of nanotechnology. Thereby combinational delivery may contribute to the synthesis of effective and novel nano-drug delivery systems and enhance their capability within pharmaceutical applications.

Stimulation of advanced research

This study can provide novel avenues for the preparation and characterization of pH-responsive LPHNPs for the co-delivery of antibiotics and natural compounds for their potential applications in pH-responsive nanosystems formulation for several antibiotic classes and other natural, bioactive compounds.

1.8 Overview of the dissertation

The research is presented in the following chapters:

Chapter One - Introduction:

This chapter provides a concise background of the study including the burden of infectious diseases on the healthcare sector, limitations of conventional antibiotic therapy and the threat of a post-antibiotic era. It further discusses the advantages of pH-responsive nanodrug delivery systems and their role in combatting antibiotic resistance. This is followed by the aims, novelty and significance of the study and concludes with an overview of the dissertation.

Chapter Two - Literature Review:

This chapter provides an overview of the burden of ID on the healthcare system and antibiotic therapy limitations that have led to the development of bacterial resistance. It also provides an overview of lipid-polymer hybrid nanoparticle characteristics, preparation and characterization technique, outlines the use of pH-responsive NDDS as a strategy to influence the treatment of infectious diseases and concludes with an overview of vancomycin as a model drug.

Chapter Three - Submitted manuscript:

This chapter is a first author article that was submitted to an international ISI journal i.e. International Journal of Pharmaceutics (Impact factor of 4.845). The chapter is presented in the required format of the journal. It describes the synthesis of novel VCM-GAPAH lipid-polymer hybrid nanoparticles. It also highlights the *in vitro* cytotoxicity evaluation, haemolytic study, formulation of the pH-responsive LPHNPs (VCM-GAPAH-LPHNPs) for targeted delivery of VCM and characterization of its physical and antibacterial properties via *in silico* studies and *in vitro* studies.

Chapter Four – Conclusions:

This chapter describes the conclusions reached in achieving the study aim, objectives, outlines the significance of the findings and provides future recommendations for further scientific research into antibiotic and natural compound co-delivery of pH-responsive lipid-polymer hybrid nanoparticles.

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Chapter Two – Literature review

2.1 Introduction

This chapter provides an overview of the burden of infectious diseases on the healthcare system and limitations of current antibiotic therapy that have led to the development of bacterial resistance. It also provides an overview of the characteristics of lipid-polymer hybrid nanoparticles, preparation and characterization techniques, outlines the use of pH-responsive NDDS as a strategy to influence the treatment of infectious diseases and concludes with an overview of VCM as a model drug.

2.2 The burden of infectious diseases on the healthcare sector and the limitations of antibiotics

Worldwide, infectious diseases pose a significant threat despite the advancement of scientific research (1). The burden of infectious diseases on the healthcare system is related to an increase in morbidity and mortality rates (Figure 1) with lower respiratory infections, lung cancer and tuberculosis (TB) being among the top ten foremost causes of death in developing and developed countries (2). Currently, lower respiratory tract infections are the leading cause of death in Africa since 2010, whereas, in South Africa, TB remains the leading cause of death (Figure 2) (3).

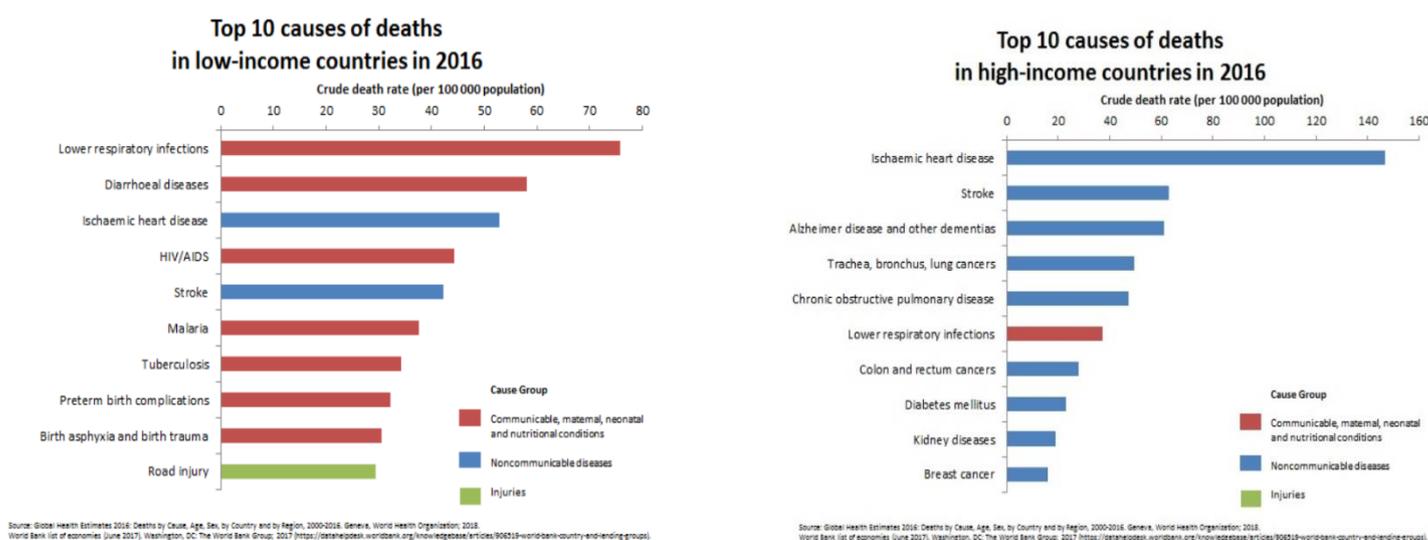


Figure 1. Leading causes of death in developing and developed countries (2).

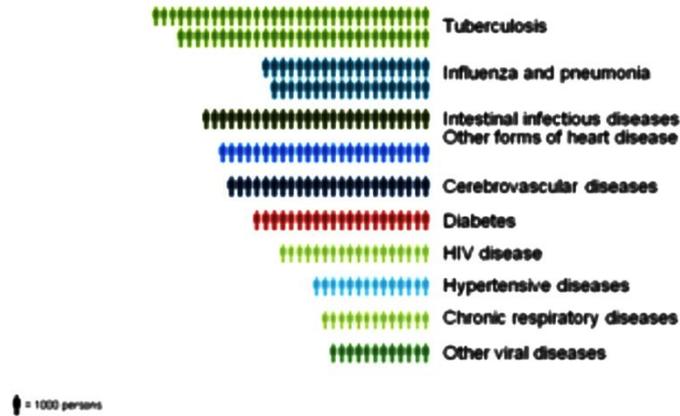


Figure 2. Leading causes of death in South Africa (4).

In 1928, the first antibiotic known as penicillin was discovered by bacteriologist, Alexander Fleming (5). In 1945, penicillin was used to treat bacterial infections (6) and thus marked the beginning of a new era of antibiotics. The discovery of conventional antibiotics contributed to the prevention and treatment of infectious diseases (7). Antibiotics are classified according to their mechanism of actions, such as cell wall synthesis inhibitors, protein synthesis inhibitors, cell membrane permeability inhibitors and nucleic acid synthesis inhibitors (Figure 3) (8). Antibiotics are produced in several dosage forms such as tablets, capsules (9), emulsions, gels (10), suppositories, creams and ointments (11).

Despite the ability to prevent and treat infectious diseases, several limitations are associated with conventional antibiotics, such as inadequate drug concentration at infection sites, harmful side effects, decreased cellular absorption and solubility, non-sustained drug release, poor pharmacokinetic profiles and poor patient compliance (12-14). Such limitations, associated with the incorrect use and unlimited use of antibiotics, has led to the failure of infectious diseases therapy, the emergence of bacterial infections and bacterial resistance (Figure 4) (14).

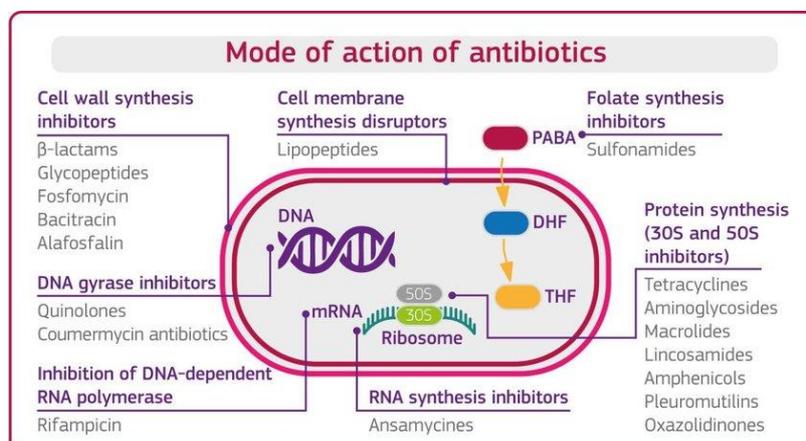


Figure 3. Commonly used antibiotics mechanism of actions (15).

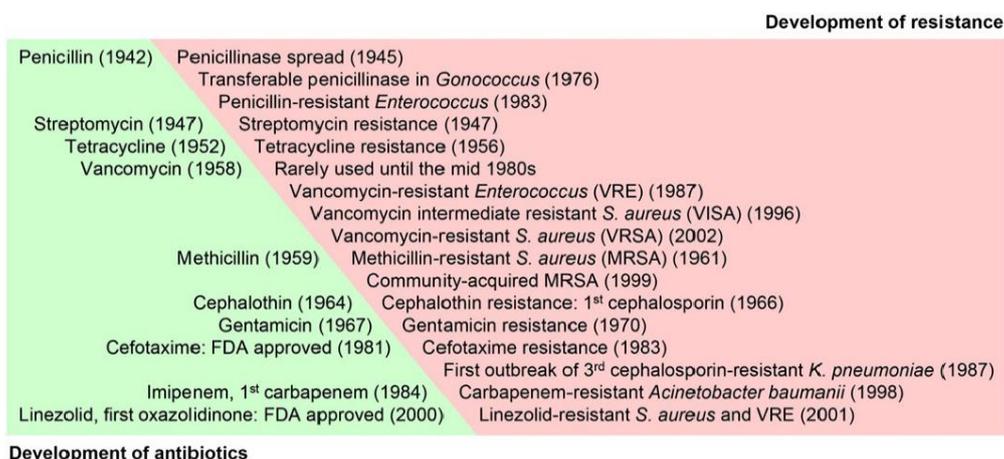


Figure 4. History of antimicrobial development and resistance of bacteria (16).

Bacteria acquire antibiotic resistance intrinsically or by horizontal gene transfer (17). Multidrug-resistant and extreme drug-resistant (XDR) bacterial strains are resistant to multiple antibiotic classes. Such untreatable bacterial infections are associated with ineffective and highly toxic antibiotic therapy (18, 19). Among universal bacterial pathogens, MRSA has developed resistance to methicillin and is responsible for skin and soft skin infections resulting in increased morbidity and mortality rates (19, 20).

Despite the increase in antimicrobial resistance, the rate of development of new antimicrobials against MDR organisms is deteriorating (Figure 5) (20). The primary reasons include the massive cost involved in the production of new chemical entities, low return on investment and lengthy drug approval procedures. Thus, novel approaches have been investigated in order to enhance the delivery of conventional drug dosage forms and to restore efficiency and effectiveness (21, 22), such as individualizing antibiotic treatment, therapeutic drug monitoring and sustained and targeted drug delivery (14, 23, 24).

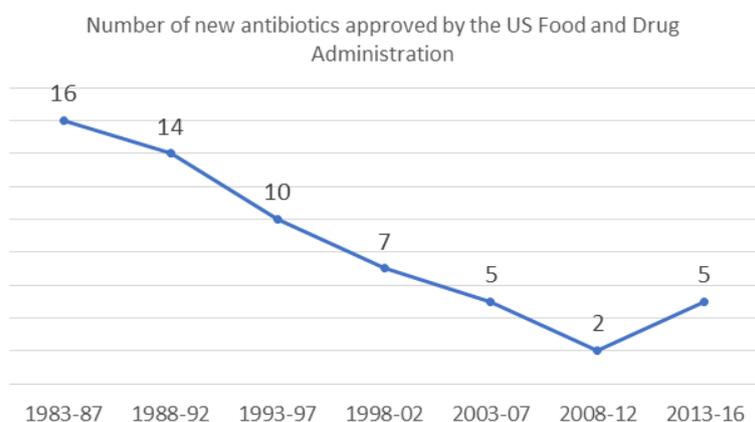


Figure 5. Declining number of new antibiotics (25).

2.3 Nanoengineered drug delivery systems

Nanotechnology is defined as the manipulation of atoms and molecules conducted on a nanoscale, used within scientific fields (26, 27). Nanotechnology within the field of medicine involves the synthesis and application of nanosized substances or compounds designed to maximize the outcome of therapy (28). Nanoparticles possess unique chemico-physical properties including, their subcellular size, large surface area to mass ratio, enhanced interactions between pathogen and host cells/tissues and ability to be modified structurally and functionally (29, 30).

Nano drug delivery systems possess the ability to improve drug stability, increase drug absorption within target tissues and enhance localization, thus enhancing drug efficacy (31). Examples of NDDS, illustrated in Figure 6, encapsulating the various drug classes include liposomes (32), PNPs (33), dendrimers (34), micelles (35), lipid polymer hybrids (36), solid lipid nanoparticles (37) and nanostructured lipid carriers (38). Table 1 illustrates several types of nanocarriers entrapping antibiotics for the treatment of different bacterial infections. Nanosystems are applied in several routes for antibiotic administration such as oral, intravenous, inhalation, topical and transcutaneous (39).

NDDS have several advantages over conventional antibiotics, which include targeted and sustained drug release at infection sites (40), selective targeting of tissues and cells (41, 42), improved cellular absorption and solubility (29, 42), improved drug stability, synergistic effects via co-delivery of antimicrobials (41, 43), enhanced patient compliance (42), minimized side effects (44) and potentiation of antibacterial activity (32). Due to the emergence of antibiotic resistant bacteria possessing evolved resistance mechanisms, innovative NDDS need to be developed to overcome antimicrobial resistance (45, 46). The various mechanisms by which nanosystems have the potential to overcome antimicrobial resistance include, increased concentration at the target site of infection, high entrapment efficiency of hydrophilic or lipophilic drugs, decreased dosage, protection of encapsulated drugs from bacterial enzymatic inactivation, increased uptake or decreased efflux, physical damage of the plasma membrane, increased removal of cytoplasmic fluid and anti-biofilm efficacy (47-49). Hence, NDDS can

overcome antibiotic limitations and offer a promising approach to combatting bacterial resistant pathogens (14).

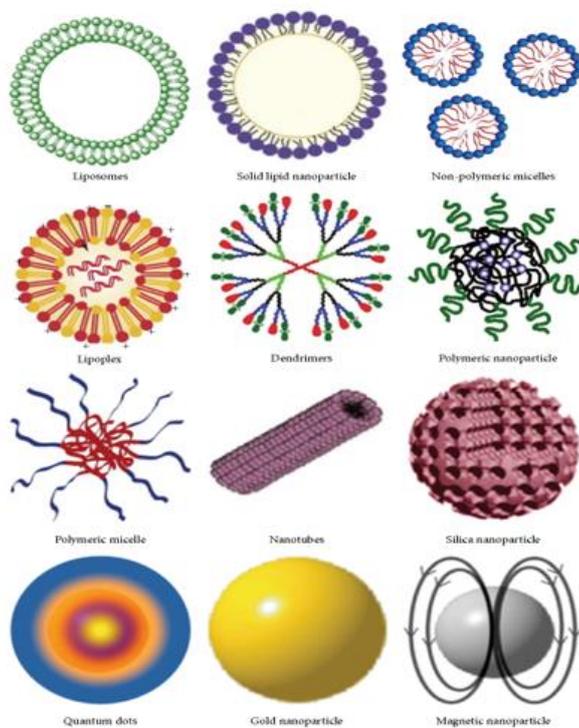


Figure 6. Examples of nanosystems reported for drug delivery (50).

Table 1. Examples of nanosystems reported for antibiotics.

Nanosystem	Encapsulated antibiotic	Targeted bacteria	Main findings	Reference
Polymeric nanoparticles	Clindamycin	MRSA	- Enhanced antibacterial activity - Enhanced wound healing	(51)
Liposomes	VCM	SA and MRSA	- Sustained release - Enhanced <i>in vitro</i> antibacterial activity	(32)
Dendrimers	Ciprofloxacin	SA and <i>Escherichia coli</i>	- Co-administration of dendrimers reduced the required effective dose of the drug - Synergistic antibacterial activity	(52)
Micelles	VCM	SA and MRSA	- Sustained drug release - Enhanced <i>in vitro</i> and <i>in vivo</i> antibacterial activity	(14)
Solid lipid nanoparticles	Meropenem	<i>Escherichia coli</i>	- Sustained release - Enhanced antibacterial activity	(13)
Nano emulsions	Amoxicillin	<i>Helicobacter pylori</i>	- Sustained drug release - Enhanced drug localization at the site of infection.	(53)
Carbon nanotubes	Ciprofloxacin	<i>Escherichia coli</i>	- Enhanced antibacterial activity	(54)
Polymersomes	VCM	SA and MRSA	- Sustained drug release - Enhanced <i>in vitro</i> and <i>in vivo</i> antibacterial activity	(55)

2.4 Lipid-polymer hybrid nanoparticles

Merging the structural components of liposomes and PNPs formulates LPHNPs (56). LPHNPs combine the potential advantages of liposomes and PNPs (56). Hybrid nanoparticles may be produced in different morphologies such as core shell and matrix LPHNPs (57). The core may be entrapped in single/multiple layers of the lipid on the polymer core material that may provide the site for surface functionalization with different targeting ligands and receptors to potentiate desired characteristics of LPHNPs (58). Natural or synthetic lipids such as, glycerol and its derivatives are amphiphilic molecules as well as convenient and inexpensive materials (59). Natural or synthetic polymers are stable, convenient and inexpensive materials for the production of numerous unique nanoparticle constructs with great potential applications in the medical field (60, 61). Additionally, polymers can also show antimicrobial properties (60, 62,

63). Such polymers demonstrate enhanced efficacy, reduced toxicity, minimal environmental hazards and provide higher resistance against MDR bacteria. Hence, the development of dynamic and non-toxic antimicrobial lipid-polymer hybrid nanoparticles are recommended for the treatment infectious diseases (64, 65). Well known synthetic polymers include polylactides (PLA), polyglycolides (PGA), an example of a copolymer is polylactide co-glycolides (PLGA) (58) and natural polymers include chitosan (66-68), alginate, albumin and gelatin (69). Figure 7 depicts the structure of LPHNPs.

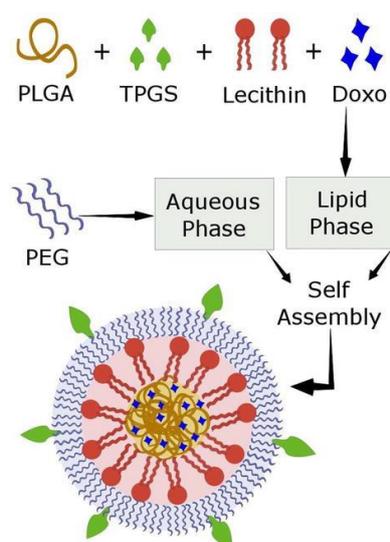


Figure 7. Lipid-polymer hybrid nanoparticle structure (70).

2.4.1 Main types of lipid-polymer hybrid nanoparticles

2.4.1.1 Monolithic lipid-polymer hybrid nanoparticles

Monolithic or mixed lipid polymer hybrids are made up of a polymeric matrix in which the lipid molecules are dispersed throughout (69). The monolithic LPHNPs are structurally comprised of copolymers and lipids as shown in Figure 8 (69). The applications include localized drug delivery, tissue engineering and cancer immunotherapy (70).

2.4.1.2 Biomimetic lipid-polymer nanoparticles

These nanoparticles are coated with red blood cell and also known as erythrocyte membrane-camouflaged PNPs (71). Lipid bilayer-coated nanoparticles are prepared by extrusion of erythrocytes and drug entrapment is within PNPs (71). These lipid-polymer-based

nanoparticles are widely applied in the fields of bioimaging, gene therapy and tissue engineering (72).

2.4.1.3 Polymer-caged liposomal nanoparticles

These are stable systems formed when polymers are anchored on the surface of liposomes (73). Their applications in the medical field include bioimaging and tissue engineering (74).

2.4.1.4 Core shell lipid-polymer hybrid nanoparticles

There are two types of core shell LPHNPs namely, polymer core-lipid shell nanoparticles (75) and hollow core-lipid-polymer-lipid nanoparticles (76). Such systems have a multi layered structure comprising of a polymeric nucleus, lipid-PEG and lipids in the outermost layers acting as the shell (77). Their applications include immunology kits and biosensors for magnifying biomolecular identification (78). Figure 8 depicts the structural image of the four main types of lipid-polymer hybrid nanoparticles.

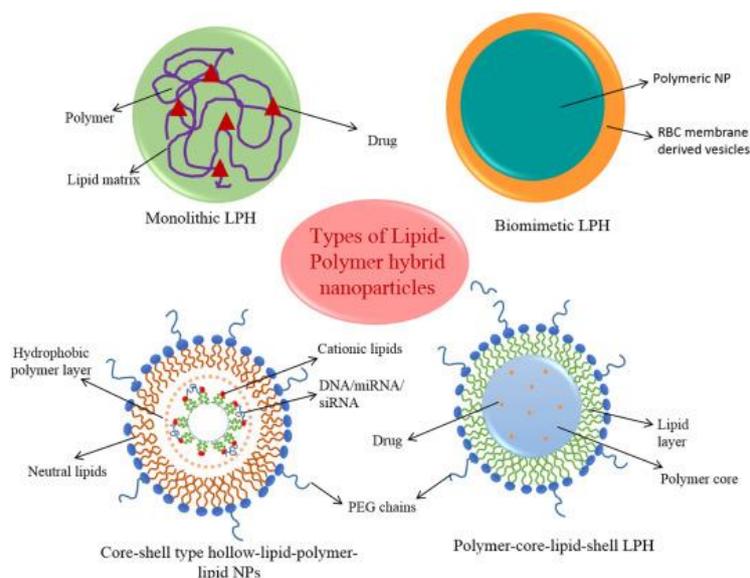


Figure 8. Four main types of LPHNPs (79).

2.4.2 Main techniques of lipid-polymer hybrid nanoparticle preparation

Several methods have been applied to produce LPHNPs from lipids and preformed polymers, including solvent evaporation (80), nanoprecipitation (81), emulsification (82), dialysis and salting out. Also, several methods have been applied to produce PNPs of LPHNPs from the

polymerization of monomers, such as emulsion, micro-emulsion, mini-emulsion and interfacial polymerization (58, 83).

Solvent evaporation was the first method developed to prepare LPHNPs (84). Using the method, shown in Figure 9, emulsions can be formulated by firstly, dissolving in volatile organic solvents. Then the emulsion is transformed into a nanoparticle suspension via evaporation of the solvent from the polymer. Two processes have been applied for the formulation of emulsions: 1) oil-in-water (single emulsions) and 2) (water-in-oil)-in-water (double emulsions). High-speed homogenization or ultra-sonication takes place followed by solvent evaporation and thereafter magnetic stirring at room temperature. LPHNPs are centrifuged and washed to remove unwanted substances (84-86).

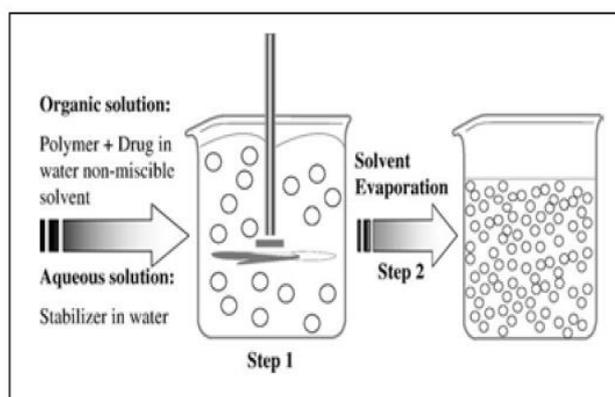


Figure 9. Solvent evaporation technique (58).

Micro-emulsion polymerization is a novel and effective approach for nanoparticle preparation (87). Particle size and the number of chains per particle are considerably lower in micro-emulsion polymerization as compared to emulsion polymerization (58, 88, 89). A water-miscible initiator is immersed into the aqueous phase of a thermodynamic micro-emulsion. Thereafter, the surfactant and all initiator molecules within the system are used up by the polymer particles. The reaction mixture consisting of swollen polymer micelles with dissolved monomers result in formed microdroplets (90, 91). Figure 10 shows the micro-emulsion technique.

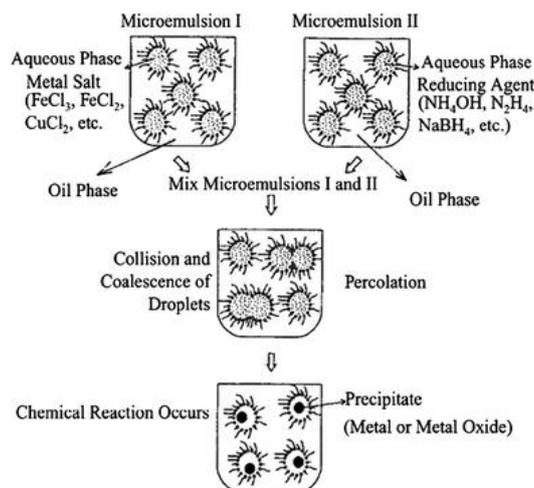


Figure 10. Micro-emulsion technique (92).

2.4.3 Lipid-polymer hybrid nanoparticle characterization

Several methods are applied for the characterization of LPHNPs in order to explore their potential application, stability and kinetic profiles. LPHNPs are characterized in terms of particle size, PDI, surface charge/ZP, morphology and other parameters depending on LPHNPs applications (93, 94).

2.4.3.1 Particle size and size distribution

Dynamic light scattering (DLS) is a technique used to determine size distribution of submicron particles or polymers in a formulation. The DLS technique is most frequently used for determining particle size and PDI as it is quick, practical and manageable (95-98). The instrument used to measure particle size and PDI is a Zetasizer. The rate at which the intensity of the scattered light fluctuates depends on the size of the particles (95).

2.4.3.2 Surface charge

The ZP of nanoparticles is a crucial indicator of the stability of LPHNPs during storage. A high positive or negative surface charge indicates high electrostatic repulsion, thus, avoiding the formation of LPHNP agglomerates; hence improved stability of the nanosystem is maintained. Analytical instruments based on the DLS technique is used for ZP measurement (99-102).

2.4.3.3 Morphology

Scanning electron microscopy (SEM) produces images of a sample by scanning the surface of a sample using a beam of electrons. The instrument used is a scanning electron microscope.

Electrons interact with the sample providing data about surface morphology and sample composition. SEM technique is advantageous for nanoparticle size and shape characterization due to easy preparation and fast image acquisition (99, 102). Similarly, transmission electron microscopy (TEM) produces images of a sample whereby an accelerated beam of electrons penetrates through a sample. The instrument used is a transmission electron microscope. Electrons interact with the sample providing data on structure and morphology. TEM is a standard method for measuring nanoparticle size, size distribution and morphology (101). Amorphous particles usually demonstrate a spherical shape, small nanoparticles form clusters and various shapes like rods and fibers can also be identified on the range of 1-1000 nm (100, 103).

2.4.3.4 Entrapment efficiency

Drug entrapment efficiency (EE) is the concentration of drug entrapped within the matrix of LPHNPs. The unentrapped or free drug is separated from the LPHNPs via ultra-centrifugation and ultra-filtration methods, followed by quantification of the free drug using UV or HPLC analysis (104-106).

2.4.3.5 Thermal profiles

Analytical instruments based on differential scanning calorimetry (DSC) techniques are used to analyze the crystallization form and thermal profile of LPHNPs (107). This technique uses a differential scanning calorimeter instrument. DSC analysis can measure melting temperature, glass transition temperature, reaction energy, crystallinity, precipitation energy and temperature (105).

2.4.3.6 In vitro drug release

Drug release studies are commonly carried out using a dynamic dialysis bag method to investigate the release behaviour of drug loaded LPHNPs, followed by quantification of the drug concentration using a reliable analytical method, such as UV or HPLC (108). Factors such as particle size (surface area) of LPHNPs can influence drug release behaviour. Smaller particles have a larger surface area to volume ratio, hence the drug associated with small nanoparticles would be situated at/near the surface resulting in a faster drug release (107, 108). Several other characterization analyses can be done, such as antibacterial studies (98) and haemolysis assays (109) for antibacterial research, cytotoxicity studies (84) for anticancer research, as well as histological evaluations and radiographic examinations for bone and tissue

engineering studies (110). Hence the type of characterization assessment depends on the application of the LPHNPs.

2.4.4 Lipid-polymer hybrid nanoparticles for antibiotic delivery

Antibiotic-loaded LPHNPs have emerged as one of the promising formulations in antibiotic therapy against infections (111). Advantages associated with LPHNPs used for antibiotic delivery are their stability during storage, easy preparation techniques and functionalization, controlled antibiotic release, improved biocompatibility and enhanced circulation time (112). The subcellular size of LPHNPs allows them to effectively penetrate the target site and release the antibiotic locally, enhancing localization and antibiotic concentration at the infection site (113). Different antibiotics can be incorporated into LPHNPs for different routes of administration such as, oral, topical, transdermal, ocular and intravenous (114, 115).

2.4.5 Responsive lipid-polymer hybrid nanoparticles for antibiotic delivery

First generation nanoparticles were developed and approved more than 15 years ago. Second generation nanocarriers advanced this field by achieving long blood circulation and passive targeting (116, 117). While third generation nanocarriers aimed at achieving molecular recognition and active targeting, hence the development of stimuli-responsive nanoparticles (116, 117). Recent scientific advancements focus on developing stimuli-responsive nanosystems that release the drugs only when triggered by pH, enzymes, temperature, electric field and magnetic field (118-122). The stimuli-responsive drug delivery strategy can provide targeted drug release, improved tissue and cellular internalization and drug accumulation at targeted sites (123). This can result in enhanced drug stability, bioavailability and therapeutic efficacy (124).

Despite the advancements of LPHNPs application incorporating different drug classes via different routes of administration such as oral (125-127), topical (128, 129), intravenous (130, 131), few studies have been reported for drug delivery via responsive LPHNPs, but with great potential. Li *et al.* reported enzyme-responsive lipid-polymer hybrid vesicles for bacterial-strain-selective delivery system for antibiotics (128), Moreno *et al.* reported pH-responsive LPHNPs for bacterial cell wall targeted delivery of antibiotics (129) and Michalak *et al.* reported antibiotic-loaded temperature-responsive LPHNPs against SA (130). Advantages of responsive LPHNPs include enhanced systemic release, enhanced internalization, improved bioavailability, minimal side effects, stabilization and protection of the antibiotic (114).

2.5 pH-Responsive LPHNPs

Current investigations of pH-responsive drug delivery systems have received much attention (132). As certain organs, cellular and tissue compartments exhibit different pH values, such as the blood, lysosomes, endosomes and gastrointestinal tract (mouth, stomach, duodenum, colon) (132). Diseases such as cancer (120), TB (133), H1N1 influenza virus (134), Alzheimer's disease (135), chronic lung diseases (136) and infections (bacterial, viral, fungal and parasitic) present with an acidic pH change different from the physiological pH of 7.4.

The human body maintains a slightly alkaline pH of 7.2, but certain conditions such as hypoxemia, whereby there is low oxygen in the blood leading to an oxygen deficiency in the tissues, otherwise known as hypoxia, can ultimately result in bodily acidosis and inflammation. Decreased oxygen intake/decreased pH (acidic pH) allows bacteria, viruses and cancer cells to thrive (137, 138). During an infection, serum lactate dehydrogenase enzyme level increases, anaerobic conditions develop and lactate production increases from pyruvate (139). Increased lactate levels elevate the production and accumulation of proinflammatory cytokines and oxygen reactive species (ROS) resulting in oxidative stress (139, 140). Lactate production continues to increment as the acidic environment increases (139, 140). This indicator can potentially trigger the design of pH-responsive nanosystems for targeted drug release (140-142). Hence, stimuli-responsive nanosystems, specifically pH-responsive drug delivery systems, can be more promising drug delivery systems than conventional NDDS. Table 2 summarizes the studies reported on pH-responsive drug delivery of LPHNPs.

At acidic conditions, pH-responsive nanoformulations operate via two mechanisms of action:

- 1) Protonation: pH-Responsive materials possess ionizable groups that remain unprotonated, while in acidic pH, these groups undergo protonation and a reversal of surface charge, leading to conformational variations followed by drug release.
- 2) Hydrolysis of labile acid bonds: pH-responsive materials can form an acid-labile bond with a drug that undergoes hydrolysis at acidic conditions, resulting in a targeted antibiotic release (121, 141, 143).

Some bacterial infections, such as SA and *E. coli*, are acidic and produce acetic acid and lactic acid under oxygen deprived conditions. Therefore, the pH-responsive approach is advantageous in reduced exposure of the antibiotic to non-infected sites, improving targeted drug delivery, controlled drug release, enhancing antibiotic localization at acidic infection sites,

which can enhance the antibacterial activity, ultimately preventing the development of bacterial resistance (129). Thus, pH-sensitive nanocarriers can impact infectious disease therapy by improving antibiotic protection at the acidic infection site, improving biofilm penetration, enhancing antibacterial activity and enhancing targeted release, thus enhancing the antibacterial activity and preventing bacterial resistance development (143-145).

Table 2. Examples of pH-responsive lipid-polymer hybrid nanoparticle delivery systems that have been tested against different target pathogens/diseases.

Lipid used	Polymer/s used	Encapsulated drug	Mechanism of action of the drug	Targeted pathogen or disease	Main findings	Reference
Bovine serum albumin	PLGA	Doxorubicin	Nucleic acid synthesis inhibitor	Multi drug resistant cancer	Enhanced drug intracellular bioavailability, improved cytotoxic activity.	(146)
Soya lecithin	PLA	Norfloxacin	Nucleic acid synthesis inhibitor	SA and <i>Pseudomonas aeruginosa</i>	Enhanced drug delivery, in vitro antibacterial activity.	(147)
Glyceryl tripalmitate	Eudragit RS100 and chitosan	Vancomycin	Cell wall synthesis inhibitor	SA and MRSA	Sustained drug release, in vitro antibacterial studies, showed better activity compared to the free drug.	(148)
Bovine serum albumin	Chitosan	Thymol	Nucleic acid synthesis inhibitor	<i>Escherichia coli</i>	Enhanced antibacterial activity, sustained drug release in simulated gastro-intestinal pH condition.	(149)
Oleic acid	Polyacrylic acid	Fusidic acid	Protein synthesis inhibitor	SA and MRSA	Controlled drug release of the drug at acidic pH conditions.	(150)

2.6. Vancomycin as a model drug for antibiotic delivery

Vancomycin is a tricyclic glycopeptide antibiotic (Figure 14), which is the ‘last resort’ drug against MRSA infections. Vancomycin functions by forming a complex with the D-Ala-D-Ala terminals, thereby inhibiting the function of peptidoglycan synthetase enzyme and subsequently preventing elongation of peptidoglycan matrix and cell wall synthesis (151-153). Intravenous administration of VCM is associated with severe side effects, such as thrombophlebitis (154), neutropenia, nephrotoxicity (155), ototoxicity, thrombocytopenia and most commonly red man or red neck syndrome (156).

2.6.1 Vancomycin nano delivery systems

Vancomycin is a large hydrophilic molecule, which accounts for its low bioavailability, cellular absorption and penetration (157). The bactericidal activity of VCM is aimed at attaching to the bacterial cell wall rather than to a protein target, initially VCM was immune to resistance (158). However, the emergence of two complex resistance mechanisms involving a multi-enzyme pathway compromised the efficacy of the drug (159). Resistance to the drug involves breakdown of D-Ala-D-Ala terminals and its replacement with D-Ala-D-lac or D-Ala-D-Ser regions to which VCM has low affinity (158, 160). Vancomycin resistance is a progressing healthcare issue that has led to the failure of VCM treatment and increased minimum inhibitory concentration (MIC) (161). Vancomycin NDDS has shown to be highly effective as compared to VCM conventional dosage forms. The liposomal-VCM formulation had enhanced antibacterial performance and sustained circulation time (162, 163), while VCM-PNPs had enhanced antibacterial activity and biocompatibility with tissues (14). VCM-LPHNs showed sustained drug release and antimicrobial activity (36). Omolo *et al.* reported VCM-loaded polymersomes that enhanced its anti-MRSA activity (55).

On the other hand, pH-responsive VCM-PNPs have been reported for enhancing VCM targeted delivery (129, 164). Kalhapure *et al.* (2017) (164) reported PNPs with size of 220.57 ± 5.9 nm and ZP of 21.9 ± 0.9 mV showed higher antibacterial activity, compared to Moreno *et al.* (2012) (129) who reported PNPs with size of 196.0 ± 7.8 nm and ZP of 2.3 ± 1.0 mV. Hence, a more positive ZP could lead to improved antibacterial activity. Kalhapure *et al.* (2017) also reported acid cleavable lipids for pH-responsive VCM-SLNs formulation; their results showed enhanced antimicrobial activity against MRSA and SA (165). Makhathini *et al.* (2020) (166) and Sonawane *et al.* (2020) (14) reported pH-responsive VCM-micelles with results showing enhanced antibacterial activity against MRSA and a faster drug release profile at pH 6.0

compared to physiological pH of 7.4. Sonawane *et al.* (2020) (14) reported micelles with size of 130.33 ± 7.36 nm and ZP of -4.33 ± 0.55 mV, which showed superior antibacterial activity than the VCM-micelles reported by Makhathini *et al.* (2020) (166) with size of 84.16 ± 0.184 nm and ZP of -42.6 ± 1.98 mV. Hence smaller particle size and a less negative ZP could lead to better antibacterial activity.

Wu *et al.* (2020) reported pH-responsive VCM-carbon dots against staphylococcal biofilms, which showed enhanced VCM penetration and killing of non-extracellular-polymeric-substance producing staphylococcal strains (167). Osman *et al.* (2019) reported pH-responsive VCM-nanostructured lipid carriers against SA and MRSA; their results indicated enhanced antibacterial activity (38). Zhang *et al.* (2020) (168) reported VCM-hybrid magnetic nanoparticles formulation; which resulted in enhanced antimicrobial activity against SA and *E. coli*. Xie *et al.* (2020) (169) reported pH-responsive VCM-silver nanoparticles formulation; their results showed enhanced antibacterial activity against MRSA and *E. coli*. Salih *et al.* (2020) (170) reported pH-responsive VCM-nanovesicles formulation; which demonstrated enhanced antibacterial activity against SA and MRSA. Thus, pH-responsive nanosystems for VCM targeted delivery can enhance VCM activity, overcome the limitations of VCM conventional dosage forms and combat bacterial resistance.

2.7 Conclusion

This chapter highlighted the potential of nano delivery systems to improve the treatment of bacterial infections and combat antimicrobial resistance. Nanocarriers that are pH responsive are emerging as a significant strategy to potentiate the performance of nanocarriers for overcoming the restrictions of antibiotic therapy and curbing the evolution of a post-antibiotic era. Furthermore, the chapter also demonstrated that vancomycin is an ideal model drug being widely used to develop novel nanosystems to treat bacterial infections. Novel pH-responsive systems of vancomycin will make a significant impact in the field of nanoantibiotics.

2.8 References

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Chapter Three

Submitted manuscript

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Chapter Three – Submitted manuscript

3.1 Introduction

The following research paper was submitted to the International Journal of Pharmaceutics which has an impact factor of 4.845 from work done during this study.

Miss Yajna Jaglal contributed to the project design and characterization of the novel nanoformulation. In addition, Miss Jaglal was responsible for the analysis and interpretation of all data, wrote the first draft of the paper and undertook all corrections. Mrs Nawras Osman assisted with formula optimization and characterization and *in vitro* drug release studies. Dr Calvin Omolo contributed to conceptualisation, design and formulation of the drug delivery system and supervision of the characterization and application studies and thesis and abstract editing. Dr Chunderika Mocktar supervised the *in vitro* antibacterial activity studies. Dr Nikita Devnarain conducted the cytotoxicity studies, *in silico* studies and editing. Prof Thirumala Govender served as supervisor and was responsible for project conceptualization, thesis and abstract editing and overall supervision of the study.

3.2 Submitted manuscript

Formulation of pH-responsive lipid-polymer hybrid nanoparticles for co-delivery and enhancement of the antibacterial activity of vancomycin and 18 β -glycyrrhetic acid

Yajna Jaglal¹, Nawras Osman^{1,2}, Calvin A. Omolo^{1,3}, Chunderika Mocktar¹, Nikita Devnarain¹ and Thirumala Govender^{1*}

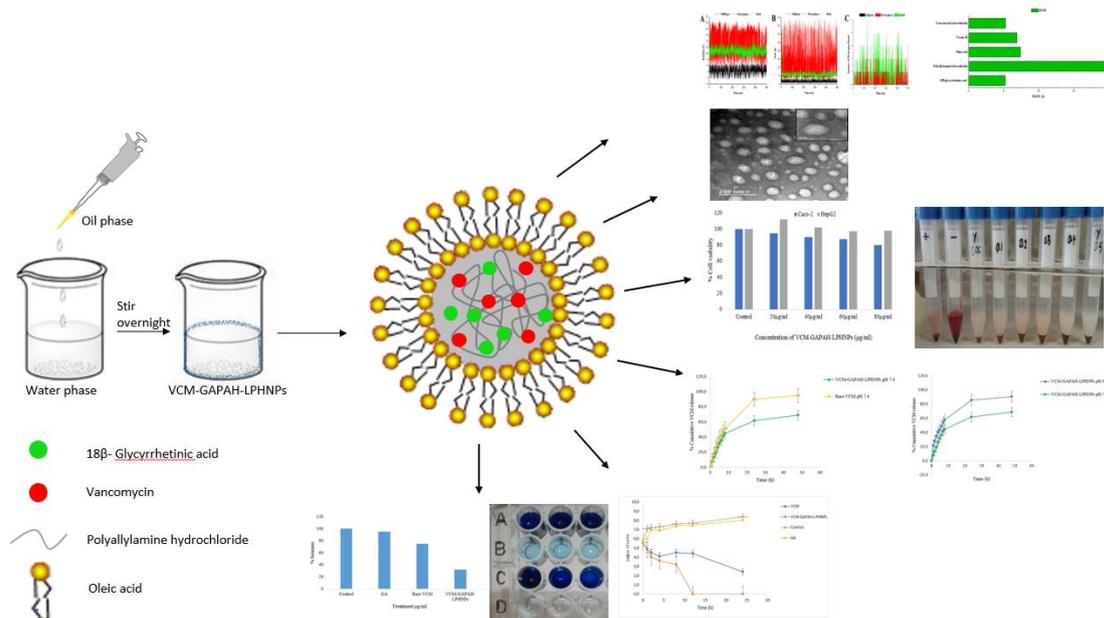
¹ Discipline of Pharmaceutical Sciences, School of Health Sciences, University of KwaZulu-Natal, Private Bag X 54001, Durban, 4000, South Africa.

² Department of Pharmaceutics, Faculty of Pharmacy, University of Gezira, Wad Medani, Sudan

³ United States International University-Africa, School of Pharmacy and Health Sciences, Department of Pharmaceutics and Pharmacy Practice, P. O. Box 14634-00800, Nairobi, Kenya.

* Correspondence: govenderth@ukzn.ac.za; Tel.: +27 31 260 7357, Fax: +27 31 260 7792.

Graphical abstract



Abstract: Despite advancement in the control and therapeutics of infectious diseases, antimicrobial resistance remains a global burden. Hence there is a need for novel strategies that improve and potentiate available antibiotics to prevent a regress to a pre-antibiotic era. This study aimed to co-deliver vancomycin (VCM) and 18 β -glycyrrhetic acid (GA) via pH-responsive lipid-polymer hybrid nanoparticles (VCM-GAPAH-LPHNPs) to explore its potential for enhanced activity and targeted delivery of VCM. The stability of VCM-GAPAH-LPHNPs were supported by *in silico* studies. Biosafe VCM-GAPAH-LPHNPs were prepared using the microemulsion technique. VCM-GAPAH-LPHNPs demonstrated size, polydispersity index, zeta potential and encapsulation efficiency of 198.4 ± 0.302 nm, 0.255 ± 0.003 , -3.8 ± 0.335 mV and 69.46 ± 2.52 %, respectively. *In vitro* drug release studies revealed that VCM-GAPAH-LPHNPs had sustained and faster release at acidic conditions compared to bare VCM. VCM-GAPAH-LPHNPs also demonstrated greater *in vitro* antibacterial potential against methicillin-resistant *Staphylococcus aureus* by 16-fold, when compared to the bare drug. Additionally, the time-killing assay indicated the ability of VCM-GAPAH-LPHNPs to eliminate 75 % of MRSA in less than 12 h. Furthermore, crystal violet assay confirmed VCM-GAPAH-LPHNPs potential to eliminate biofilms. Therefore, these novel LPHNPs may serve as promising nanocarriers for enhancing antibiotic drug delivery and antibacterial activity.

Keywords: pH-responsive; lipid-polymer hybrid nanoparticles; vancomycin; methicillin-resistant *Staphylococcus aureus*; 18 β - glycyrrhetic acid

1. Introduction

Globally, infectious diseases primarily due to bacterial infections, are considered one of the main contributors towards increasing morbidity and mortality rates (1, 2). The WHO reported that more than 17 million deaths are caused by infectious diseases per year (3). Recent data indicate that by 2050, more individuals will die from bacterial diseases than any other disease, including cancer (4). Moreover, by then, approximately 4 million deaths are predicted in Africa due to antimicrobial resistance, with a possibility of higher death rates if appropriate measures are not undertaken (5). Hence, urgent interventions are required for the development of novel and innovative strategies to curtail current and emerging antibiotic-resistant pathogenic bacterial strains that are no longer sensitive to conventional antibiotics (6, 7).

Current traditional antibiotic dosage forms possess various limitations, including suboptimal drug concentration at infection sites, high exposure to healthy cells, frequent administration of high doses and prolonged therapy (8-10). This results in sub-optimal drug delivery and activity, increased adverse effects, the toxicity of cells/tissues and poor patient compliance, thus contributing to poor outcomes and development of antibiotic resistance (11, 12). Hence, the engineering, formulation and application of novel drug delivery systems (NDDS) are explored (9) as an advanced strategy to suppress the limitations related to conventional dosage forms that contribute to antimicrobial resistance (13, 14).

Novel drug delivery systems have showcased their potential in addressing the disadvantages related to conventional antibiotics by enhancing drug delivery at target sites of infection due to their subcellular size, biocompatibility with host cells/tissues and sizeable surface area to mass ratio (9, 15, 16). Lipid-polymer hybrid nanoparticles (LPHNPs) are one of the promising systems that are currently being employed to deliver drugs efficiently (17, 18). LPHNPs combine the mechanical advantages of polymeric nanoparticles and prowess of lipidic systems (17, 19). They are core-shell self-assembled nanosystems composed of a hydrophobic lipid core and a polymeric shell (20). Flexibility in the formulation of LPHNPs allows the formulation of the system that has programmable ability to respond to stimuli (21, 22). This can be achieved by focusing on the biofunctional property of the nanoparticle matrix in order to introduce targeted drug delivery and enhanced activity (23). The advancement in the field of nanotherapeutics has encouraged the development of pH-responsive nanosystems for effective and efficient antibiotic delivery (24-26). However, there are few reports on employing pH-responsive LPHNPs for antibiotic delivery. Therefore using this strategy, systems can be

designed to have programmable destabilization and release the drug due to changes in pH that are synonymous with bacterial infection sites (27, 28).

Plants contain rich sources of bioactive phytochemical compounds that are active against a wide spectrum of activity, 18 β -glycyrrhetic acid (GA) is such a compound (29, 30). It is a pentacyclic triterpenoid metabolite of a hydrolyzed product of glycyrrhizic acid from the *Glycyrrhiza glabra* (licorice) plant (31). This bioactive compound has been reported to have anti-allergic (32), antibacterial, antiviral, antitumor and anti-inflammatory properties (33, 34). Due to the antibacterial activity of GA, it can be explored by co-delivery with other antibiotics in the market for synergistic action and augmentation of antibacterial drugs in the market (33, 35).

Surveys from the literature show combinational therapy of antibiotics in the market and bioactive compounds from natural sources have shown to enhance antibacterial activity (36). However, currently, there are no reports of GA being co-delivered with antimicrobial agents in the market in a nanosystem to target bacterial infectious diseases. Therefore, the combination of GA and antibiotics can further be explored by co-loading in a stimuli-responsive nanosystem for targeted and further enhancement of antimicrobial activity (37). Such a system will also contribute to the advancement of the pharmaceutical field.

There are several reports of co-delivering GA with cancer and inflammatory drugs loaded in nanosystems for the treatment of cancer diseases (37-40). However, from our search of the literature, such a formulation for co-delivery of GA and antibiotics has not been reported before. We herein report vancomycin (VCM) and GA loaded LPHNPs that was coined from PAH and OA. This is the first report of LPHNPs for co-delivery of a hydrophilic drug and hydrophobic antibacterial agent in a pH-responsive drug delivery system for targeting bacteria. Therefore, the aim of this study was to formulate a multifunctional LPHNP that is pH-responsive and co-loaded with two antibacterial agents. We envisage enhanced antibacterial activity due to the synergistic activity of the two loaded antibacterial agents, in addition to pH-responsive targeted delivery of VCM and GA as a result of protonation and deprotonation of the OA and PAH. In basic media, OA will deprotonate becoming negatively charged and it will electrostatically combine with the positive amines of PAH, while in acidic media both OA and PAH protonate and both molecules become positively charged thus repelling each other. Hence, the disintegration of the LPHNPs system will lead to an increase in drug release. Moreover, protonation and deprotonation will lead to surface switching LPHNPs from negative

in basic media to positive in acidic media. To the best of our knowledge, such a multifunctional LPHNP system has never been reported before.

Furthermore, no study has reported the pH-responsive co-delivery of VCM and GA nanoparticles to target bacteria. The advantages of the co-delivery of both hydrophilic and hydrophobic agents for combinational therapy, pH-responsive release for targeted delivery of VCM and GA and surface charge switching of the systems to target the negatively charged bacterial cell membrane can potentiate antibacterial effects of the system by acting in different mechanisms. The formulation and evaluation of this novel multifunctional LPHNP are reported in this paper.

2. Materials and Methods

2.1 Materials

18 β - Glycyrrhetic acid, polyallylamine hydrochloride (PAH), Tween 80, VCM hydrochloride, oleic acid (OA), dialysis tubing cellulose membrane, Mueller Hinton broth 2 (MHB) were obtained from Sigma-Aldrich (USA). Mueller Hinton Agar (MHA) and Nutrient Broth were obtained from Biolab Inc. (South Africa). Milli-Q purified water was obtained from an Elix[®] water purification system Millipore Corp. (USA). Sheep blood was purchased from United Scientific SA cc. (South Africa). Bacterial strains used were *Staphylococcus aureus* (Rosenbach) (ATCC[®]BAA-1683) (MRSA) and *Staphylococcus aureus* (ATCC 25922) (SA). The CACO-2 and HepG2 cell lines were purchased from ATCC.

2.2 Molecular dynamics of components of oil phase, water phase and both phases

To study the stability of the oil phase and water phase independently and then the assembly of the oil and water phase to form a complex, the AMBER suite was used to simulate the oil phase components, GA and OA, then separately the water phase components including PAH, VCM and Tween 80 and finally both oil and water phase components. Molecular dynamic simulations represent an all-encompassing toolkit that delves into the atomic arrangement within molecules, thus providing novel perspectives on the structural movements of molecular systems. The molecular dynamics simulations were carried out using the PMEMD engine of the AMBER software package with GPU acceleration (41). Gasteiger charges were used to charge compounds and ANTECHAMBER was used to create atomic partial charges using the General AMBER Force Field and Restrained Electrostatic Potential (RESP) methods. The LEAP module of AMBER 14 was employed to combine, neutralize and solvate all systems via

the addition of hydrogen atoms and chloride and sodium ions and suspended the systems in an orthorhombic box of TIP3P water molecules such that all atoms were within 10 Å of the box edges. The protein residues were renumbered due to missing residues in the initial crystal structure. An initial minimization was performed for 2500 steps with a restraint potential of 10 kcal.mol⁻¹Å⁻² to the solutes and for 500 steps of steepest descent followed by 500 steps of the conjugate gradient. This was followed by 1000 steps of full minimization Langevin thermostat, with a collision frequency of 1.0 ps⁻¹ with harmonic restraint of 5 kcal.mol⁻¹ Å⁻² on the solutes, applied during the gradual heating up of the systems to a temperature of 300 K in the canonical ensemble for 50 ps. This was followed by 50 ps of density equilibration in isothermal–isobaric ensemble and a final 500 ps equilibration at 300 K, 1 bar pressure and a coupling constant of 2 ps. The simulations were performed for 500 ns using classical molecular dynamics with a time step of 2 fs, with the frame being recorded at every 500 steps of simulation. All the bond lengths involving hydrogen atoms were constrained using the SHAKE algorithm (42). All the MD simulations were carried out using the GPU Amber 14 software package (43).

2.3 Preparation of VCM-GAPAH-LPHNPs

A previously reported micro-emulsion technique was used to prepare the VCM and GA loaded pH-responsive LPHNPs (44). Briefly, the oil phase, consisting of OA (40 mg) and GA (10 mg) was dissolved in ethanol (2 ml). While the aqueous phase was prepared by dissolving PAH (10 mg), VCM (10 mg) and Tween 80 (20 mg) in 20 ml distilled water. The oil phase was added in a dropwise manner to the aqueous phase under stirring overnight at 500 rpm.

2.4 Size, Polydispersity Index (PDI), Zeta Potential (ZP) and Morphology

The size, PDI and ZP of VCM-GAPAH-LPHNPs were determined by the dynamic light scattering (DLS) technique using a Zetasizer Nano ZS90 instrument (Malvern Instruments Ltd., UK) at 25 °C. An aliquot of LPHNPs was appropriately diluted with phosphate buffers (pH 7.4, 6 and 4.5) to obtain concentrations that were within the system's sensitivity range, the experiments were performed in triplicate (44, 45).

The morphology of VCM-GAPAH-LPHNPs was determined by transmission electron microscopy (TEM). Samples were suitably diluted, stained with 1 % uranyl acetate (UA) solution, air-dried and visualized using a TEM (JOEL JEM-1010, Japan) operating at an accelerating voltage of 100 kV (46).

2.5 Entrapment efficiency (EE%) and drug loading (DL%)

The EE% and DL% of VCM-GAPAH-LPHNPs were determined by an ultrafiltration method. Briefly, LPHNPs emulsion (3 ml) was inserted into Amicon[®] Ultra-4 centrifugal filter tubes (Millipore Corp., USA) with 10 kDa pore size and centrifuged at 2000 rpm at 25 °C for 20 min (14). The untrapped VCM was detected using a validated High-Pressure Liquid Chromatograph (Shimadzu, Japan) with UV detection at λ 280 nm. The mobile phase, consisting of water with 0.1 % trifluoroacetic acid (TFA) and acetonitrile (85/15 v/v), was pumped through a Nucleosil 100-5C18 column (150 mm X 4.6 mm in diameter) at a flow rate of 1 ml per min and an injection volume of 100 μ l, with a regression equation of $Y = 0.0044x - 0.0013$ and linearity coefficient (R^2) of 0.9997. EE% and DL% were calculated using equations (1) and (2) below (27, 47).

$$\text{EE (\%)} = \left(\frac{\text{Weight of VCM/GA in LPHNPs}}{\text{Weight of VCM/GA added}} \right) \times 100 \quad (1)$$

$$\text{DL (\%)} = \left(\frac{\text{Weight of VCM/GA in LPHNPs}}{\text{Total weight of nanoparticles}} \right) \times 100 \quad (2)$$

2.6 Thermal profiles

Differential scanning calorimetry (DSC) using a DSC-60 (Shimadzu, Japan) was used to determine the thermal profiles of the bare VCM, GA, physical composition and lyophilized drug-loaded LPHNPs. Briefly, each sample (2 mg) was placed in an aluminium pan and sealed. The scanning was done at a temperature range of 30 °C - 300 °C at a constant rate of 10 °C/min under a constant nitrogen flow of 10 ml/min, an empty pan was used as a reference (48).

2.7 *In vitro* drug release and release kinetics

The *in vitro* release studies were performed using a dialysis bag technique to investigate the release profile and mechanism of VCM-loaded GAPAH-LPHNPs. Dialysis bags with a pore size of 8000–14,400 Da containing the VCM-GAPAH-LPHNPs (VCM/GA 0.5 mg/ml each) were placed in PBS (40 ml) of pH 6 and 7.4 at 37 °C in a shaking incubator at 100 rpm. At 0.5, 1, 2, 3, 4, 5, 6, 7, 8, 24, 48, 72 h, the samples (3 ml) were withdrawn and replaced with equal quantities of the fresh PBS solutions of pH 6 and 7.4 to maintain a constant volume. The amount of VCM released per time interval was determined by HPLC (Shimadzu, Japan) with UV detection set at λ 280 nm, with a regression equation of $Y = 14129x + 115582$ and linearity coefficient (R^2) of 0.9071, all experiments were performed in triplicate (1, 2, 49).

Drug release kinetics were analysed using the DDSolver software program [60]. Six mathematical models, namely zero order, first order, Higuchi, Hixson-Crowel, Korsmeyer-Peppas and Weibull were used to calculate and compare the correlation coefficient (R^2) and root-mean-square error (RMSE) values.

2.8 *In vitro* cytotoxicity

The biocompatibility of VCM-GAPAH-LPHNPs was determined using the MTT assay following a previously reported method (50) on two cell lines, which included human intestinal epithelial cancer cells and human liver adenocarcinoma cells (CACO-2 and HepG2, respectively). Briefly, the cell lines were cultured and supplemented with 10 % foetal bovine serum (FBS), 1 % streptomycin and penicillin solution and 1 % L-glutamine, which were incubated at 37 °C in a humidified atmosphere of 5% carbon dioxide (CO₂). Following 80 % confluency, the cell lines were seeded into a 96-well plate. After incubation for 24 h, the culture medium containing the treatment samples (20, 40, 60 and 80 µg/ml) were replaced with fresh culture medium (100 µl per well) and MTT solution (20 µl per well). After 4 h of incubation, the culture media and MTT assay solutions were immediately removed and dimethyl sulfoxide (100 µl) was added to each well to solubilize the MTT formazan crystals. The absorbance corresponding to each well was measured at λ 570 nm (Spectrostar Nano, Germany); all experiments were performed in triplicate. Equation (3) was used to determine the percentage cell viability:

$$\text{Cell viability (\%)} = \left(\frac{\text{A570nm treated cells}}{\text{A570nm untreated cells}} \right) \times 100 \quad (3)$$

2.9 Haemolysis

Haemolytic toxicity of VCM-GAPAH-LPHNPs was determined at different concentrations (15). Briefly, sheep blood was washed thrice with an isotonic 0.1 M PBS solution (pH 7.4) by centrifugation at 3000 rpm for 5 min. For each sample, VCM-GAPAH-LPHNPs formulation was diluted with 0.1 M PBS to obtain concentrations ranging from 0.05 to 0.5 mg/ml. The red blood cell (RBC) suspension (0.2 ml) was added to 1.8 ml of each sample and incubated at 37 °C for 30 min. Thereafter the samples were centrifuged at 3000 rpm for 10 min. The supernatant of each sample was then collected and analysed for haemoglobin released by UV spectrophotometry at λ 416 nm. To obtain 0 % and 100 % haemolysis, 0.2 ml of RBC

suspension was added to 1.8 ml of PBS and distilled water, respectively, as a control. The degree of haemolysis was calculated using equation (4):

$$\text{Haemolysis (\%)} = \left(\frac{\text{ABS} - \text{ABS}_0}{\text{ABS}_{100} - \text{ABS}_0} \right) \times 100 \% \quad (4)$$

where ABS_{100} and ABS_0 are the absorbances of the samples at 100% and 0% haemolysis, respectively.

2.10 Stability studies

The stability of VCM-GAPAH-LPHNPs was tested using DLS for 0,30,60 and 90 days of storage, at 4 °C and room temperature (51).

2.11 *In vitro* antibacterial activity and fractional inhibitory concentration (FIC) index

The minimum inhibitory concentration (MIC) values of bare VCM, GA, VCM-GAPAH-LPHNPs and blank-GAPAH-LPHNPs were determined against SA and MRSA (pH 6 and 7.4) using the broth dilution method. Briefly, bacteria culture was grown in Nutrient Broth at 37°C in a shaking incubator (Labcon, USA) at 100 rpm for 18 h. Bacterial cultures were suitably diluted to achieve a concentration equivalent to 0.5 McFarland's Standard using a DEN-1B McFarland densitometer (Latvia). This was again diluted to 1:150 with sterile distilled water to achieve colony forming units per ml (CFU/ml) of 5×10^5 . All tested samples were serially diluted in MHB (pH 6 and 7.4) and incubated with the diluted bacterial cultures at 37°C in the shaking incubator at 100 rpm for a total of 96 h. At 24, 48 and 72 h, tested samples (5 µl) were spotted onto MHA and incubated at 37°C for 24 h. The bare drug formulation was used as the positive control while the blank formulation of LPHNPs was used as a negative control. All experiments were performed in triplicate (52-54).

The antibacterial effect of bare VCM and GA in combination with VCM-GAPAH-LPHNPs against SA and MRSA was determined using the cumulative FIC of bare VCM and GA based on the Loewe additivity zero-interaction theory. The FIC index was calculated using equations 5 and 6 and the FIC index is shown in (Table 1).

$$\text{FIC formula: } \Sigma \text{ FIC} = \text{FIC of agent A} + \text{FIC of agent B}$$

$$\text{FIC of agent A} = \frac{\text{MIC of agent A in combination with agent B}}{\text{MIC of agent A alone}} \quad (5)$$

$$\text{FIC of agent B} = \frac{\text{MIC of agent B in combination with agent A}}{\text{MIC of agent B alone}} \quad (6)$$

Agent A: Bare VCM

Agent B: GA

Agents (A-B) in combination: VCM-GAPAH-LPHNPs

Table 1. FIC index.

FIC index	Interpretation
$\Sigma \text{ FIC} \leq 0.5$	Synergism
$> 0.5 \Sigma \text{ FIC} \leq 1$	Additive
$> 1 \Sigma \text{ FIC} < 2$	Indifference
$\Sigma \text{ FIC} \geq 2$	Antagonism

Synergism combination indicates that a lower concentration of agents A and B is required in order to produce the same effect as agent A and agent B alone. While additive combination indicates that the same concentration of agents A and B are required in order to produce the same effect as agent A and agent B alone, the indifference combination indicates agent B is inactive since there was no effect of it to be added but only the effect of the active agent, which is agent A. However, antagonism combination indicates that higher concentration of agents A and B are required in order to produce the same effect as agent A and agent B alone (55-58).

2.12 Time-killing assay

The time-killing analysis of VCM-GAPAH-LPHNPs was conducted using the plate colony count method, as previously reported (59, 60). MRSA was cultured for 48 h at 37 °C in nutrient broth and a shaking incubator followed by bacterial dilution in sterile PBS to obtain

concentrations of $10^5 - 10^6$ CFU/ml (61). Thereafter bare VCM, GA, sterile water and VCM-GAPAH-LPHNPs were added to the PBS containing MRSA at a concentration of 5 times greater than MIC. The samples were placed in the shaking incubator at 37 °C. At 1, 2, 4, 8, 12 and 24 h intervals, 0.1 ml of each sample was sub-cultured on nutrient agar plates for 24 h. The number of colonies counted were converted to \log_{10} values and plotted on a graph. The experiment was performed in triplicate.

2.13 MRSA biofilm reduction

The eradication of MRSA biofilms by VCM-GAPAH-LPHNPs was determined using crystal violet (CV) assay (62). Briefly, 100 μ l of MRSA suspensions (1.5×10^8 CFU/ml) in nutrient broth was inserted into a 96-well plate and incubated for 3 days at 37 °C to form a mature biofilm. Prior to treatment, the media was removed from the wells and the wells were thoroughly washed with PBS (pH 7.4) to remove non-adherent bacterial cells. Precisely, 100 μ l of bare VCM solution (390 μ g/ml), GA solution (3,125 μ g/ml) and VCM-GAPAH-LPHNPs (390 μ g/ml) formulation were added to the wells at a concentration of 100 times greater than MIC and incubated for 12 h at 37 °C. The wells were washed with PBS for the removal of the treatments and non-adherent bacterial cells. Thereafter, the wells were fixed with methanol for 15 mins. The plate was air-dried for 0.5 h, followed by the wells being stained with 0.1 % CV solution and kept in total darkness at room temperature (25 °C) for 20 mins. After washing with distilled water, 30 % of acetic acid was added to each well. The absorbance per well was measured at λ 570 nm using equation (7) (Spectrostar Nano, Germany); all experiments were performed in triplicate.

$$\text{Percentage biomass (\%)} = \left(\frac{\text{A570nm untreated cells} - \text{A570nm treated cells}}{\text{A570nm untreated cells}} \right) \times 100 \quad (7)$$

3. Results and Discussion

3.1 Structural Dynamics

Molecular dynamics simulations were applied to three systems, i.e. components of the oil phase, water phase and both phases together for 500 ns to reveal the stability of the nanosystem using virtual conditions that mimic experimental conditions, expounding on the structural dynamics of each system.

Root-mean-square deviation (RMSD) calculations confirmed that with both phases separately and together, all components of each system remained stably bound to each other, without losing interaction (Figure 1A). The oil phase simulation showed slight changes in conformation

as OA wraps around GA, while the water phase simulation showed the convergence of PAH, Tween 80 and VCM to form a complex where Tween 80 held the complex together. However, the components of the water phase merged with gaps in the complex, which were accommodated by OA and GA when all components of both phases were simulated together. Altogether, the oil and water phase together formed a stable complex that remained intact throughout the simulation.

The stability and convergence of each system were confirmed by analyzing the RMSD during the simulation (Figure 1A). While the RMSD values of the oil phase system (black) remained within a 2 Å range, that of the water phase system (red) were much higher and went beyond a 12 Å range with a deviation toward the end of the simulation. The combined oil and water phase system (green) corrected for these values as the system remained within a 2 Å range and did not have any extreme deviation throughout the simulation, although higher than the oil phase system owing to a larger complex size. In addition, the radii of gyration (RoG) of all three systems (Figure 1B) indicate that when the oil phase is combined with the water phase, the system becomes more compact than that of the water phase system alone, which correlates with the small size of the nanoparticles observed in section 3.1. These data are substantiated by results of hydrogen bond analysis (Figure 1C), which confirmed a greater number of hydrogen bonds in the complex with both phases combined (green), as opposed to either of the phases separately. This may have been a contributing factor to the compactness of the entire system.

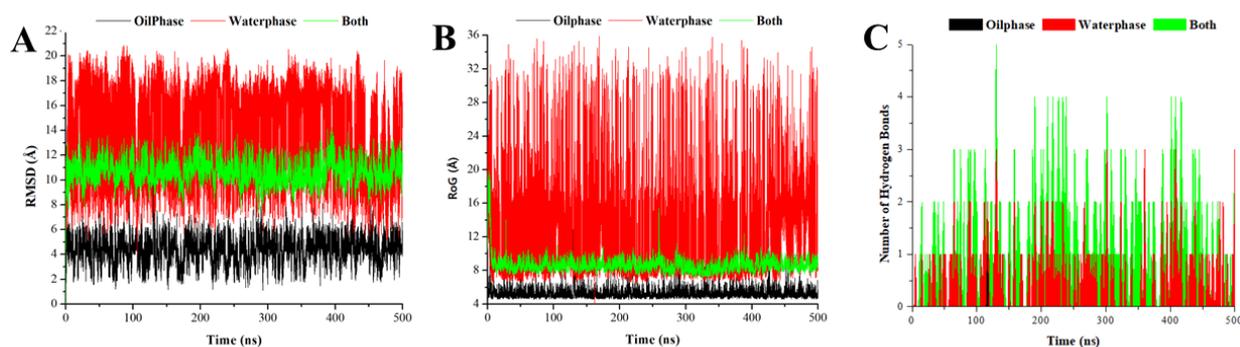


Figure 1. Post analyses of simulations of oil phase, water phase and both phases combined.

Graph A. indicates RMSD analysis, B. RoG analysis and C. hydrogen bond analysis.

Root-mean-square fluctuation (RMSF) analysis revealed energy fluctuations of each component of the 500 ns simulation of the combined oil and water phase. Figure 2 indicated that PAH fluctuated more than any other molecule in the simulation, while VCM and GA had the least amount of movement.

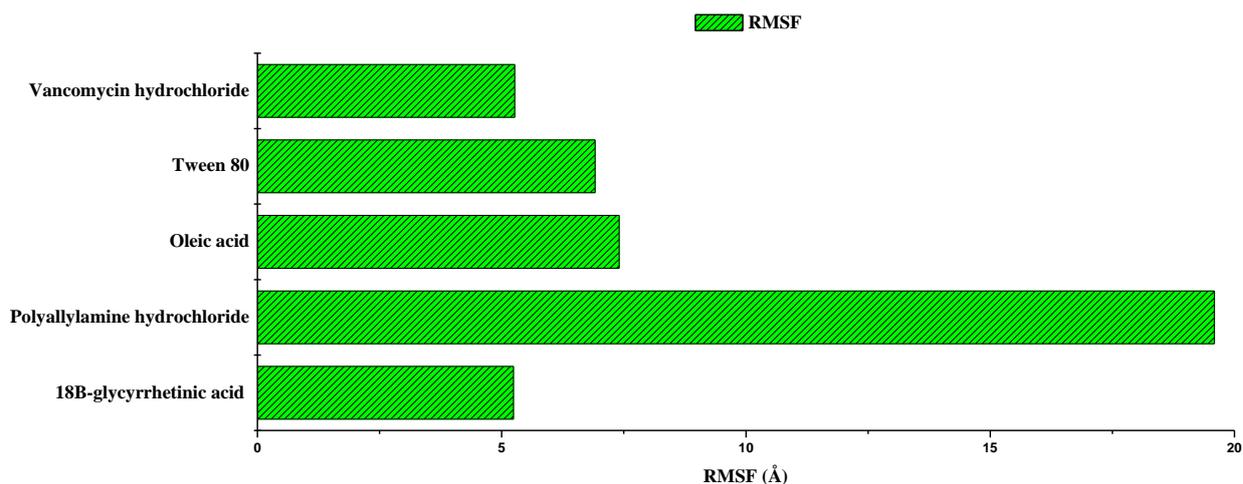


Figure 2. RMSF analysis of 500 ns simulation of all components of the combined oil and water phase system.

From these molecular dynamics analyses, it may be hypothesized that all five components of the oil and water phases bind stably and compactly to support the small size obtained from experimental data. Thus, this nanosystem has potential in the delivery of VCM to MRSA for the improvement of antibacterial activity.

3.2 Preparation of VCM-GAPAH-LPHNPs

VCM-GAPAH-LPHNPs were prepared using GA, PAH (polymer), OA (lipid), Tween 80 (surfactant) and VCM hydrochloride by micro-dilution as previously reported (63). Preliminary studies were conducted using different ratios of lipid and polymer to obtain a formulation with optimum size, PDI and ZP according to pH 7.4, 6 and 4.5 conditions with maximum VCM/GA entrapment efficiency (Table 2).

Table 2. Particle size, PDI and ZP of VCM-GAPAH-LPHNPs.

PAH: OA (mg)	pH	Particle size (nm)	PDI	ZP (mV)	EE %
1:6	7.4	323.8 ± 0.5588	0.291 ± 0.007	- 7.4 ± 1.36	
	6	353.1 ± 9.824	0.728 ± 0.035	- 0.084 ± 0.035	38.60 ± 2.84
	4.5	360.1 ± 2.967	0.288 ± 0.003	0.639 ± 0.484	
1:5	7.4	209.7 ± 1.244	0.277 ± 0.002	- 6.48 ± 0.25	
	6	280.1 ± 9.223	0.311 ± 0.0032	0.169 ± 0.4	43.06 ± 0.55
	4.5	280 ± 1.45	0.243 ± 0.01	0.708 ± 0.383	
1:4	7.4	198.4 ± 0.302	0.255 ± 0.003	- 3.8 ± 0.335	
	6	374.8 ± 5.435	0.279 ± 0.054	0.44 ± 0.325	69.46 ± 2.52
	4.5	450.4 ± 1.33	0.377 ± 0.012	0.693 ± 0.44	
<i>p</i> value = 0.037					

At pH 7.4, as the polymer: lipid ratio decreased from 1:6 to 1:5, the particle size and PDI of the LPHNPs decreased from 323.8 nm to 209.7 nm and 0.29 to 0.27, respectively, whereas the ZP values changed from – 7.4 mV to – 6.48 mV, while the EE% improved from 38% to 43%. The increase in particle size relative to the increase in lipid content, may have resulted in an increased viscosity, thus causing the LPHNPs to swell (72). Alternatively, the lower polymer: lipid ratios formed more stable nanoparticles with homogenous size distributions and enhanced EE.

The VCM-GAPAH-LPHNPs optimal formulation displayed size of 198.4 ± 0.302 nm, PDI of 0.255 ± 0.003 and ZP of -3.8 ± 0.335 mV. In addition, the surface charge switched from negative at pH 7.4 (physiological pH) to positive at 6 (acidic pH). This may be due to the deprotonation of the carboxylic group (COOH) in OA, which may form an electrostatic bond with the primary amine groups of PAH, thereby neutralising the cationic effect (64). The positive charge of VCM-GAPAH-LPHNPs at acidic pH can be advantageous for the adhesion to the negatively charged bacterial membrane (65, 66).

The VCM-GAPAH-LPHNPs EE% and DL% were found to be 69.46 ± 2.52% and 13.45 ± 0.68%, respectively. This encapsulation was comparable to other VCM-loaded LPHNPs (28) and VCM-loaded nanoparticles (67, 68). Thus, enhanced entrapment efficiency may be due to the high separation of the hydrophobic VCM hydrochloride within the lipid-polymer matrix

(69). The results suggest that VCM-GAPAH-LPHNPs can effectively deliver drugs to acidic sites of infection and were further characterized.

3.3 Characterization of optimal VCM-GAPAH-LPHNPs

The TEM study of VCM-GAPAH-PNPs displayed a distinct circular shape with a size of 198 nm, which corresponded with results of the DLS technique (Figure 3). To determine pH-responsiveness, VCM-GAPAH-LPHNPs were placed in pH 7.4, 6 and 4.5 buffer solutions. As the pH changed from 7.4 to 6 and then to 4.5, the size of VCM-GAPAH-LPHNPs increased from 198.4 nm to 374.8 nm to 450.4 nm, the PDI increased from 0.255 to 0.279 to 0.377 and the ZP switched from -3.8 mV to 0.44 mV to 0.693 mV.

At pH 7.4 conditions, LPHNPs were negatively charged due to the presence of OA on the LPHNPs matrix. At low pH values, the amine group and carboxylic group remained protonated, resulting in cleavage of the ionic bonds between PAH and OA and this was displayed by the positive ZP at acidic pH conditions (70). The cleavage of ionic bonds decreased the affinity between the polymer and lipid resulting in the deconstruction of LPHNPs dispersion, thus causing an increase in their particle size. The size change observed relates to the structural transformation of LPHNPs and the negative to positive charge switching was predicted to positively affect the drug release and promote binding to the negatively (anionic) charged bacterial cell wall, thus improving antibacterial activity (69).

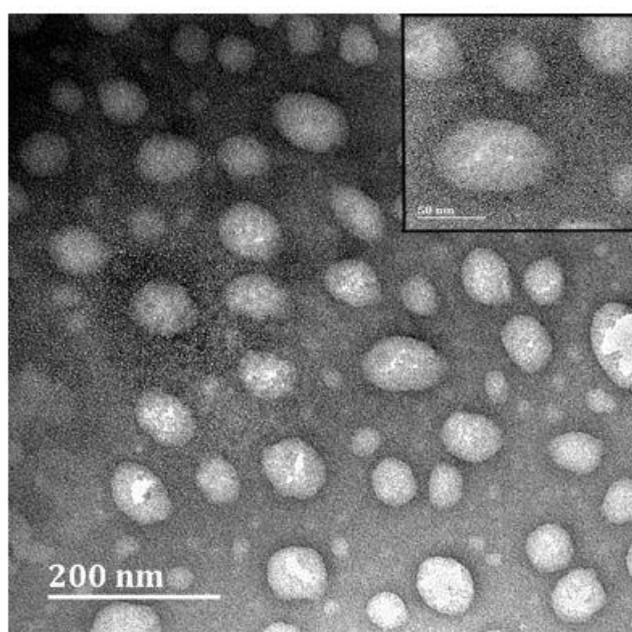


Figure 3. TEM image of VCM-GAPAH-LPHNPs.

3.4 Thermal profiles of VCM-GAPAH-LPHNPs

Thermal analysis was performed to investigate the thermal behaviour of the components of LPHNPs. The thermal profiles of the bare VCM, PAH, GA, physical composition and lyophilized VCM-GAPAH-LPHNPs were compared (Figure 4). VCM, PAH and GA peaks of were observed at 103.40 °C, 262.99 °C and 265.72 °C, respectively. The physical mixture demonstrated similar behaviour to the individual materials. The lyophilized VCM-GAPAH-LPHNPs formulation showed the disappearance of the VCM thermal peak, confirming the transformation from crystallization into its amorphous form, as it was encapsulated within the LPHNPs matrix (71).

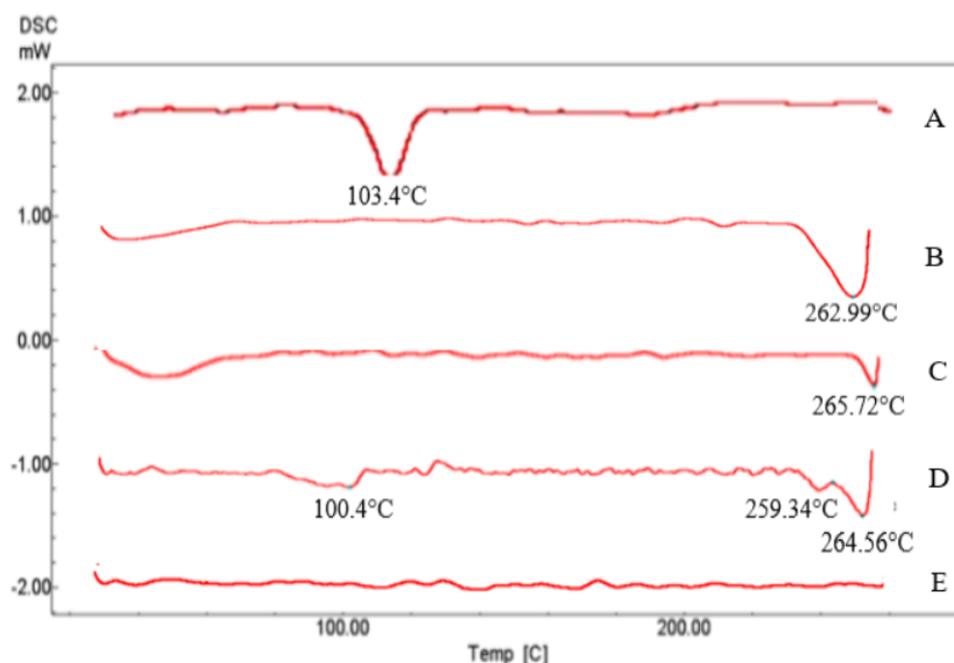


Figure 4. DSC thermogram of (A) VCM (B) PAH (C) GA (D) physical composition of VCM, PAH and GA (E) lyophilized VCM-GAPAH-LPHNPs.

3.5 *In vitro* drug release of VCM-GAPAH-LPHNPs

The *in vitro* release profile of pH-responsive VCM/GA loaded LPHNPs was investigated using the dialysis bag method at pH 7.4 and pH 6. Compared to the release of bare VCM, the release of VCM-GAPAH-LPHNPs was slower, indicating controlled release (Figure 5). The controlled release of VCM-GAPAH-LPHNPs could be attributed to the lipid-polymer matrix of VCM-GAPAH-LPHNPs, which entrap the drug for a more extended period (72). In addition, the long carbon chain length of OA due to branching could have decelerated the rate of diffusion of

VCM, resulting in slower drug release (73), which is advantageous for extended and sustained antibacterial activity (74). From the 1st hour, VCM-GAPAH-LPHNPs illustrated a significantly faster release ($p < 0.05$) at pH 6 than at pH 7.4 (Figure 6), for up to 48 h. An elevated release at acidic pH occurred due to the protonation of OA and PAH at acidic pH, which resulted in the cleavage of the ion pair bond and an increase in the size of nanoparticles due to electrostatic repulsion between PAH and OA. This resulted in the swelling of LPHNPs, causing them to burst leading to an enhanced and faster VCM release (75, 76). This pH-sensitive release activity is crucial for enhanced VCM protection at pH 7.4, enhanced drug release and bioavailability at the acidic infection sites, thus enhancing antibacterial activity (75, 77).

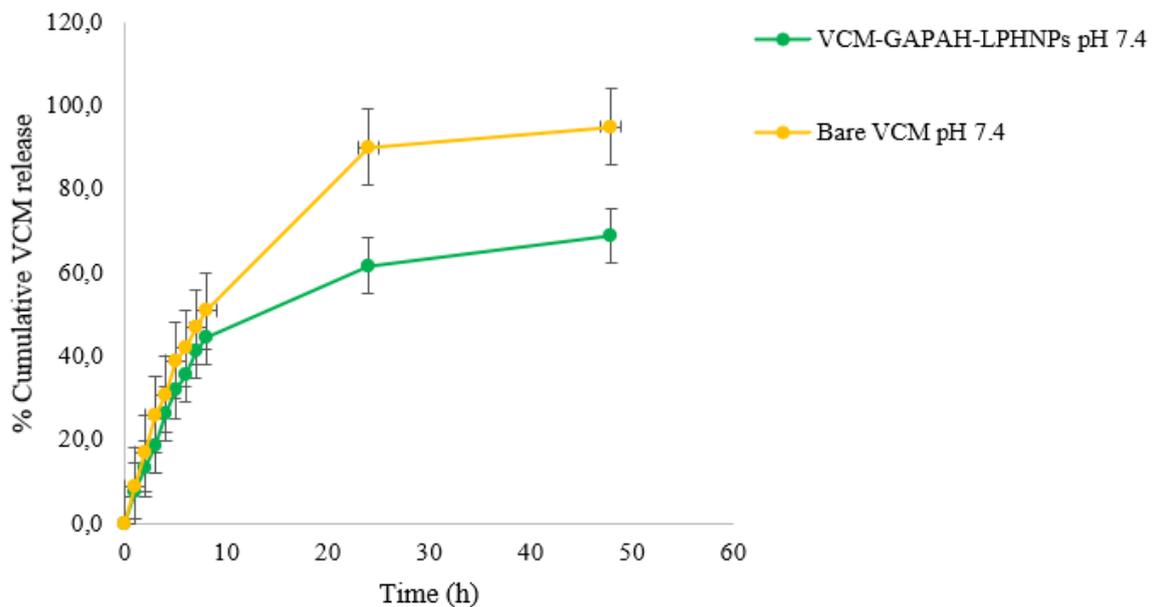


Figure 5. Drug release profiles of bare VCM and VCM-GAPAH-LPHNPs at pH 7.4 (n=3).

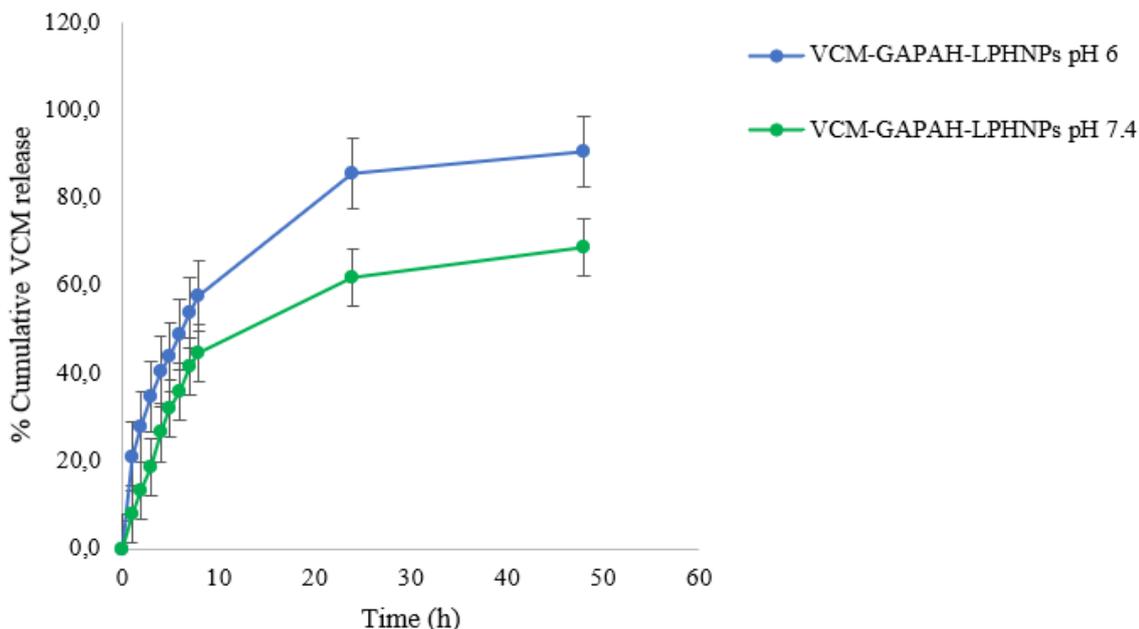


Figure 6. Effect of pH on drug release profiles of VCM-GAPAH-LPHNPs (n=3).

Regarding kinetic analysis, optimized VCM-GAPAH-LPHNPs were input into zero-order, first-order, Higuchi, Hixson-Crowell, Korsmeyer-Peppas and Weibull models (Table 3). Although the release of the VCM-GAPAH-LPHNPs was higher at pH 6, it had similar kinetic behaviour to the release at pH 7.4. The highest R^2 values were 0.945 and 0.991, while the lowest RMSE values were 4.853 and 2.189 at pH 7.4 and pH 6, respectively. Thus, the Weibull model was considered the best fit model for VCM-GAPAH-LPHNPs release at pH 7.4 and pH 6.

The Weibull release exponent (β) value is used to determine the drug release mechanism, with $\beta \leq 0.75$ indicating a Fickian diffusion and $0.75 < \beta < 1$ indicating a combined mechanism, whereas a $\beta > 1$ is associated with a collapse release mechanism. The β values for VCM-GAPAH-LPHNPs were 0.553 and 0.741 at pH 7.4 and pH 6, respectively, indicating Fickian diffusion as the release mechanism.

The Korsmeyer-Peppas release exponent (n) value is also used to determine the drug release mechanism, with $n \leq 0.43$ indicating a Fickian diffusion, $0.43 < n < 0.85$ indicating a combination of diffusion and erosion, whereas $n \geq 0.85$ suggests an erosion release mechanism. The (β) values of Weibull model were consistent with the (n) values of Korsmeyer-Peppas

model, whereas (n) values were 0.414 and 0.351 at pH 7.4 and pH 6 respectively, confirming the Fickian diffusion mechanism of VCM release from the VCM-GAPAH-LPHNPs (78, 79).

Table 3. Drug release kinetics data for VCM-GAPAH-LPHNPs.

Model	R ²		RMSE		Release exponent	
	7.4	6	7.4	6	7.4	6
Zero order	0.028	-0.764	20.578	30.870	-	-
First order	0.746	0.895	10.470	7.069	-	-
Higuchi	0.867	0.745	22.754	11.325	-	-
Hixson-Crowell	0.614	0.735	19.4	11.224	-	-
Korsmeyer-Peppas	0.893	0.945	6.8	5.582	0.414	0.351
Weibull	0.945	0.991	4.853	2.189	0.553	0.741

3.6 *In vitro* cytotoxicity

A cytotoxicity study was conducted to determine the biosafety of the formulation. The cytotoxicity of VCM-GAPAH-LPHNPs was determined using the MTT assay, performed on CACO-2 and HepG2 cells. The results showed that VCM-GAPAH-LPHNPs maintained over 75% cell viability across concentrations of 20-80 µg/ml after 24 h, indicating non-cytotoxicity (80) (Figure 7). The results from the MTT assay using VCM-GAPAH-LPHNPs indicated a high percentage of cell viability and dose-dependent trends for all cell lines. Since VCM-GAPAH-LPHNPs showed no significant cytotoxic effect on any of the cell lines, it is thus safe for use in biomedical applications.

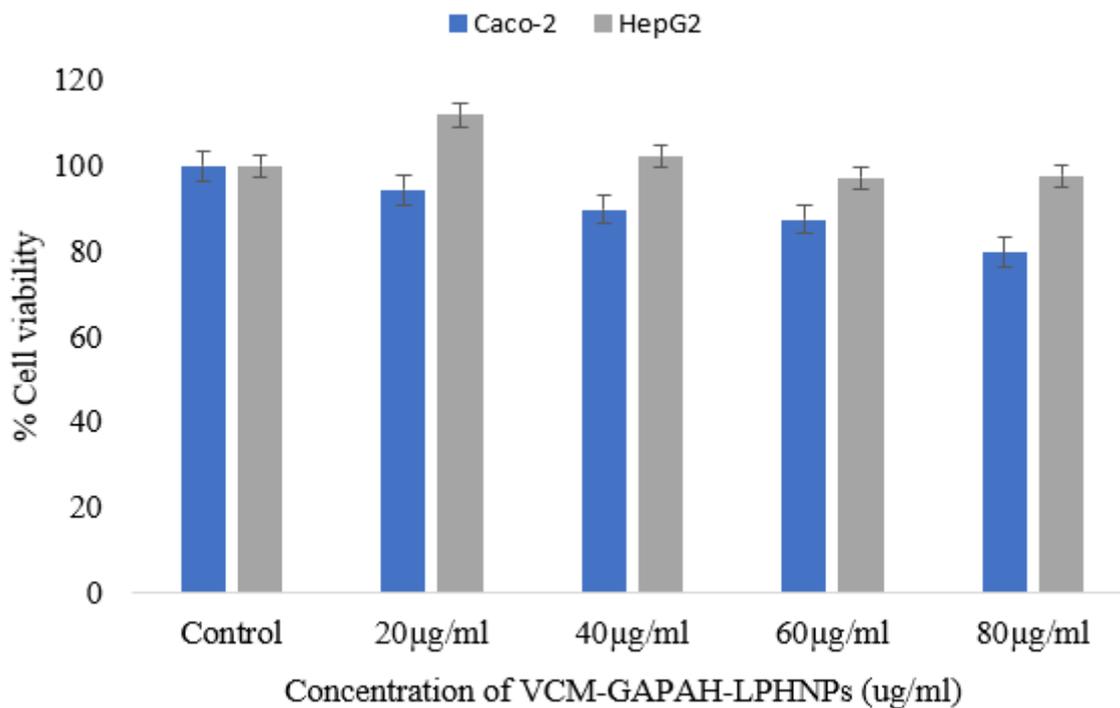


Figure 7. MTT assay of VCM-GAPAH-LPHNPs.

3.7 Haemolysis

Haemolysis of RBCs due to lipid-polymer hybrid systems may be detrimental; therefore, it is important to evaluate the safety of formulation. Haemolytic properties of VCM-GAPAH-LPHNPs were evaluated on sheep RBCs (Figure 8). VCM-GAPAH-LPHNPs were found to be non-haemolytic to RBCs in the concentration range tested (0.05 to 0.5 mg/ml). At 0.5 mg/ml, VCM-GAPAH-LPHNPs showed <1% haemolysis, indicating the non-haemolytic property to RBCs.

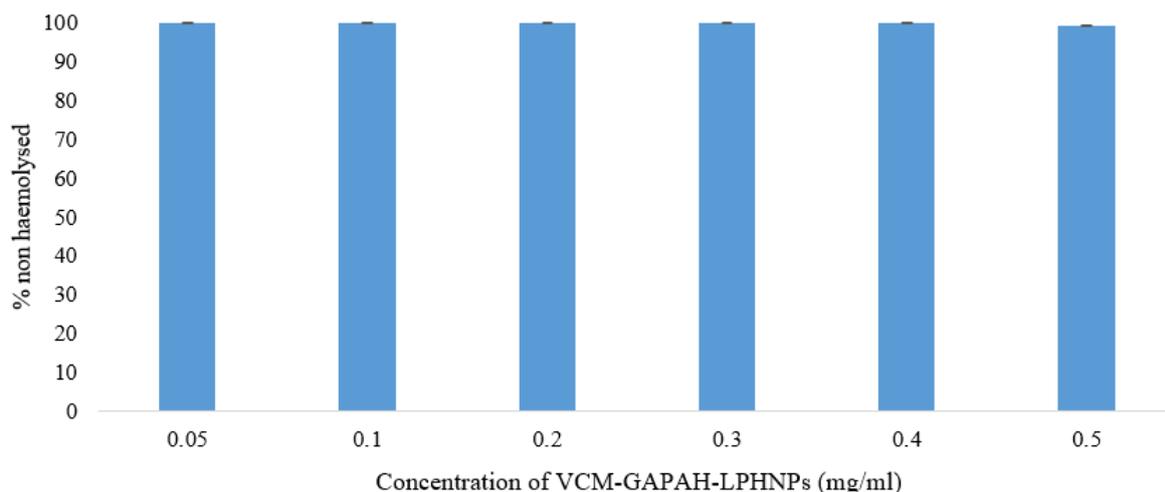


Figure 8. Haemolysis assay of VCM-GAPAH-LPHNPs.

3.8 Stability studies

The experimental results show no significant changes in particle size, PDI and ZP of VCM-GAPAH-LPHNPs throughout the 90 days of storage, at 4 °C and at room temperature – the system remained stable (Table 4).

Table 4. Effect of storage conditions and time on particle size, PDI and ZP of VCM-GAPAH-LPHNPs.

Time (days)	4 °C			Room temperature (25 °C)		
	Particle size (nm)	PDI	ZP (mV)	Particle size (nm)	PDI	ZP (mV)
0	198.4 ± 0.302	0.255 ± 0.003	-3.8 ± 0.335	198.4 ± 0.302	0.255 ± 0.003	-3.8 ± 0.335
30	198.5 ± 0.402	0.155 ± 0.002	-3.63 ± 0.88	200.5 ± 1.124	0.257 ± 0.004	-1.17 ± 0.02
60	199 ± 0.452	0.251 ± 0.041	-2.8 ± 0.55	205.2 ± 8.25	0.38 ± 0.028	-4.18 ± 2.26
90	199.9 ± 0.40	0.26 ± 0.005	-2.82 ± 0.35	206.4 ± 3.2	0.46 ± 1.33	-5.1 ± 3.14

3.9 *In vitro* antibacterial activity and FIC index

The micro broth dilution method was used to determine the MIC of VCM-GAPAH-LPHNPs at pH 7.4 and 6 against SA and MRSA (Table 5). The MIC values for bare VCM against SA and MRSA at pH 7.4 was 1.95 µg/ml and 3.9 µg/ml, respectively and increased to 3.9 µg/ml and 7.8 µg/ml, respectively, at pH 6.

The MIC values of VCM-GAPAH-LPHNPs against SA and MRSA at pH 7.4 was 7.8 µg/ml and 3.9 µg/ml respectively and decreased to 3.9 µg/ml and 0.48 µg/ml at pH 6. Thus, VCM-

GAPAH-LPHNPs had improved and sustained antibacterial activity compared to bare VCM activity against MRSA only at both pH levels. The enhanced extended antibacterial activity of VCM-GAPAH-LPHNPs, in comparison to the bare VCM, could be attributed to its subcellular size with a large surface area and the presence of GA acid for its known antibacterial properties (81, 82). Additionally, the lipophilic nature of VCM-GAPAH-LPHNPs could allow for enhanced uptake into the bacterial cell wall, thus enhancing VCM activity (83). Furthermore, the presence of OA could have enhanced VCM-GAPAH-LPHNPs activity (84).

On the other hand, at pH 7.4, GA activity against SA and MRSA was extended up to 48 h and 72 h respectively, while at pH 6, GA had activity for only 24 h against SA and MRSA. More importantly, VCM-GAPAH-LPHNPs demonstrated lower MIC values against MRSA at pH 6 than pH 7.4. At 24 h, VCM-GAPAH-LPHNPs activity was 8 times greater against MRSA at pH 6 than at pH 7.4, while at 48 h and 72 h, VCM-GAPAH-LPHNPs was 4 times greater against MRSA at pH 6 than at pH 7.4. At 96 h, VCM-GAPAH-LPHNPs was 2 times greater against MRSA at pH 6 compared to pH 7.4.

The enhanced antibacterial activity of VCM-GAPAH-LPHNPs against MRSA at acidic pH conditions may be due to VCM-GAPAH-LPHNPs protonation and cleavage of ionic bonds between PAH and OA, thus resulting in an enhanced VCM release. Furthermore, VCM-GAPAH-LPHNPs charge switching to positive in acidic conditions can enhance the binding of VCM-GAPAH-LPHNPs with the negatively charged bacterial cell wall, leading to a higher release and localization of VCM at acidic infection sites, thereby increasing its uptake into the bacterial cell wall (85, 86).

The formation of ion pairs between OA and PAH could have enhanced the diffusion of VCM and GA across the thicker peptidoglycan layer of MRSA compared to SA (87). In addition, the enhanced penetration of VCM and GA, along with OA into the lipophilic bacterial cell membrane could have contributed to the enhanced antibacterial activity of MRSA compared to SA (88).

Table 5. *In vitro* antibacterial activity at pH 7.4 and pH 6.

Time (h)	SA				MRSA			
	24	48	72	96	24	48	72	96
<i>In vitro</i> antibacterial activity at pH 7.4								
Bare VCM	1.95	3.9	3.9	7.8	3.9	7.8	15.6	15.6
VCM-GAPAH-LPHNPs	7.8	3.9	3.9	7.8	3.9	3.9	3.9	7.8
GA	125	125	NA	NA	31.25	62.5	62.5	NA
Blank GAPAH-LPHNPs	NA	NA	NA	NA	NA	NA	NA	NA
<i>In vitro</i> antibacterial activity at pH 6								
Bare VCM	3.9	3.9	7.8	15.6	7.8	15.6	15.6	31.25
VCM-GAPAH-LPHNPs	3.9	3.9	7.8	7.8	0.48	0.97	0.97	3.9
GA	125	NA	NA	NA	31.25	NA	NA	NA
Blank GAPAH-LPHNPs	NA	NA	NA	NA	NA	NA	NA	NA

NA = No activity. The values are expressed as mean (n=3).

FIC index was calculated using equations 5 and 6, to determine the combined impact of VCM and GA against SA and MRSA at pH 7.4 and 6 (Table 6). For 24, 48, 72 and 96 h against SA at pH 7.4, Σ FIC was 4.062, 1.031, 1 and 1, respectively, indicating antagonistic, indifferent and additive combinations, respectively. Whereas at pH 6, Σ FIC was 1.031, 1, 1 and 0.5 respectively, indicating indifferent, additive and synergistic combinations, respectively.

Furthermore, at pH 7.4, Σ FIC against MRSA was 1.124, 0.562, 0.312 and 0.5 respectively, indicating indifferent, additive and synergistic combinations, respectively. Lastly, Σ FIC against MRSA at pH 6, was 0.076, 0.062, 0.062 and 0.012 respectively, indicating synergistic combinations at all time intervals.

Synergy indicates that a lower concentration of VCM-GAPAH-LPHNPs is required in order to produce the same effect as VCM and GA alone. Synergistic interactions can potentially increase antibacterial efficacy and decrease toxicity. Whereas additivity indicates that the same concentration of VCM-GAPAH-LPHNPs is required in order to produce the same effect as VCM and GA alone. However, indifference is when one compound, which is GA, is inactive because there was no effect of it to be added but only the effect of the active drug, which was VCM. Lastly, antagonism indicates that a higher concentration of VCM-GAPAH-LPHNPs is required in order to produce the same effect as bare VCM and GA alone. Antagonistic

interactions can potentially decrease antibacterial efficacy and increase toxicity (55-58). Significantly, VCM-GAPAH-LPHNPs works best against MRSA at pH 6, confirming that GA and VCM had a synergistic effect at the acidic conditions.

VCM hinders bacterial cell wall synthesis by binding to the D-alanyl-D-alanine moiety of the growing peptide chain (89), whereas GA hinders bacterial cell membrane synthesis via the inhibition of genes involved in carbohydrate, amino acid and nucleic acid metabolism (35, 90). Moreover, studies have shown that GA inhibits bacterial DNA replication which could result in the inhibition of the production of bacterial toxins and enzymes (91). GA also has the ability to regulate the production of haemolysins, leukotoxins and adhesins (92, 93). Long *et al.* (2013) reported a decrease in the expression of *saeR*, *hla*, *mecA* and *sbi* genes after SA incubation with GA (94). In addition, Li *et al.* (2012) reported the downregulation of RNAlII transcript after MRSA incubation with GA (95). Hence GA can modulate virulence by synergistically inducing antibacterial effects. However, the precise mechanism underlying its activity remains unknown (94, 95).

On the other hand, the administration and exposure of VCM to human cells has several harmful effects such as nephrotoxicity, neutropenia and ototoxicity (96). However, GA can cause alkalosis, muscular paralysis and hyperkalaemia (97). By co-encapsulating VCM and GA within LPHNPs, their pharmacokinetic profiles and therapeutic indices are remarkably enhanced (98). Therefore, the synergistic combination of VCM and GA against MRSA at acidic pH can potentially increase the antibacterial efficacy, reduce cytotoxicity and prevent the emergence of antimicrobial resistance as compared to monotherapy regimens (55-57).

Table 6. FIC index against SA and MRSA at pH 7.4 and 6.

SA (pH 7.4)				
Time (h)	Bare VCM	GA	Σ FIC (Bare VCM + GA)	Interpretation
24	4	0.062	4.062	Antagonism
48	1	0.031	1.031	Indifference
72	1	-	1	Additive
96	1	-	1	Additive
SA (pH 6)				
24	1	0.031	1.031	Indifference
48	1	-	1	Additive
72	1	-	1	Additive
96	0.5	-	0.5	Synergy
MRSA (pH 7.4)				
24	1	0.124	1.124	Indifference
48	0.5	0.062	0.562	Additive
72	0.25	0.062	0.312	Synergy
96	0.5	-	0.5	Synergy
MRSA (pH 6)				
24	0.061	0.015	0.076	Synergy
48	0.062	-	0.062	Synergy
72	0.062	-	0.062	Synergy
96	0.012	-	0.012	Synergy

3.10 Time-killing assay

Time-kill analysis results of bare VCM, GA, control and VCM-GAPAH-LPHNPs at 5 times greater MIC against MRSA are illustrated in Figure 9. At 8 h interval GA eliminated 26% of bacteria, bare VCM eliminated 55% of bacteria and ultimately, VCM-GAPAH-LPHNPs eliminated 68% of bacteria. VCM-GAPAH-LPHNPs displayed spontaneous bacterial elimination with nearly 75% clearance of MRSA in <12 h with an equal concentration of bare VCM. At 12 h VCM-GAPAH-LPHNPs demonstrated 100% bacterial elimination, whereas bare VCM demonstrated 66% bacterial elimination. Hence VCM-GAPAH-LPHNPs could effectively treat MRSA infections at a faster rate.

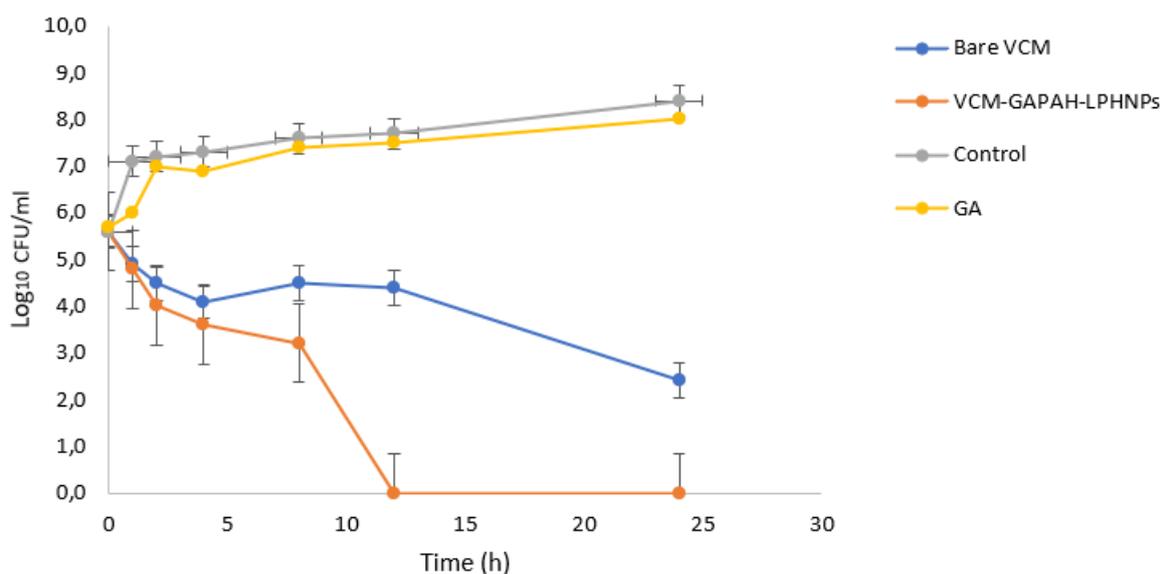


Figure 9. The killing kinetics of MRSA exposed to 5x MIC of bare VCM, VCM-GAPAH-LPHNPs, GA and sterile water (control).

3.11 MRSA biofilm elimination

The CV assay was performed to analyse the ability of VCM-GAPAH-LPHNPs to decrease MRSA biofilms by calculating the biomass percentage (Figure 10). The control (untreated biofilms) demonstrated high purple intensity (Figure 11A) resulting in a high percentage biomass of 100 %. The percentage biomass of GA, bare VCM and VCM-GAPAH-LPHNPs present were 95 %, 74 % and 31 %, respectively. Hence GA, bare VCM and VCM-GAPAH-LPHNPs eliminated 5 %, 26 % and 69 % of MRSA biofilms, respectively.

Bare VCM demonstrated a reduced purple intensity (Figure 11B), indicating a lower percentage of MRSA biomass, compared to the control. Hence a decrease in biofilms was observed when treated with bare VCM, demonstrating its considerable antibacterial ability. Notably, GA treated biofilms exhibited higher purple intensity (Figure 11C) than VCM treated biofilms, which indicated a higher percentage of MRSA biomass.

Interestingly, a substantial decrease in biofilms treated with VCM-GAPAH-LPHNPs formulation was observed, as shown in the reduction of purple intensity (Figure 11D), when compared to all other treated and untreated samples. This confirms the superior potential of VCM-GAPAH-LPHNPs to reduce MRSA biofilms.

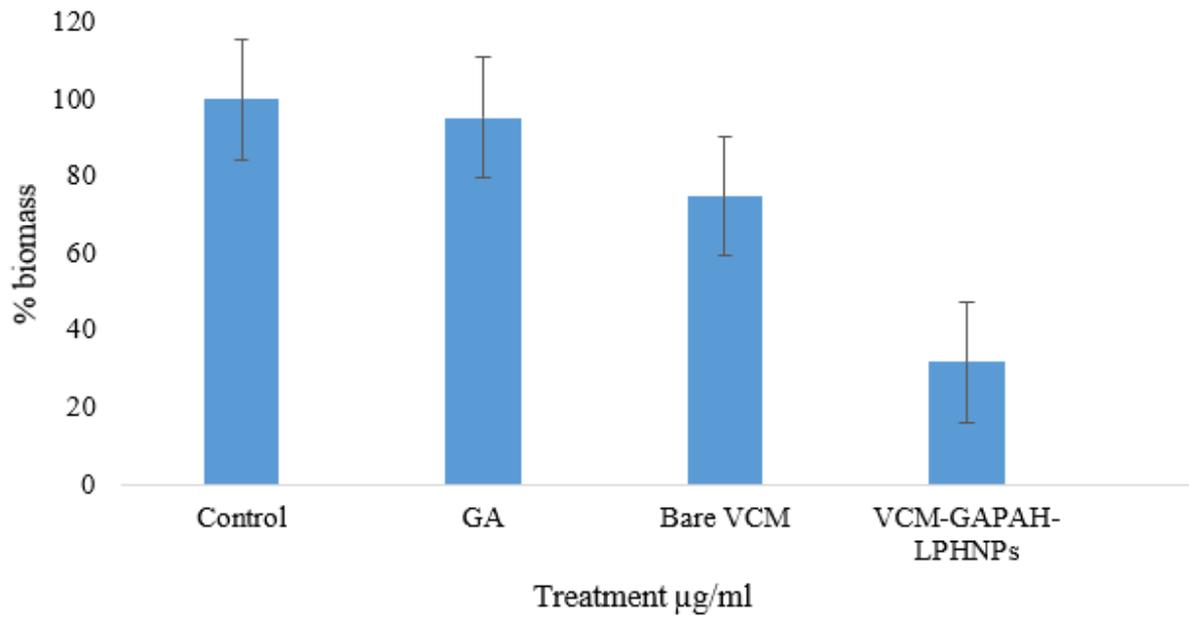


Figure 10. Percentage biomass of MRSA exposed to 100x MIC of control (untreated biofilms), GA, bare VCM and VCM-GAPAH-LPHNPs.

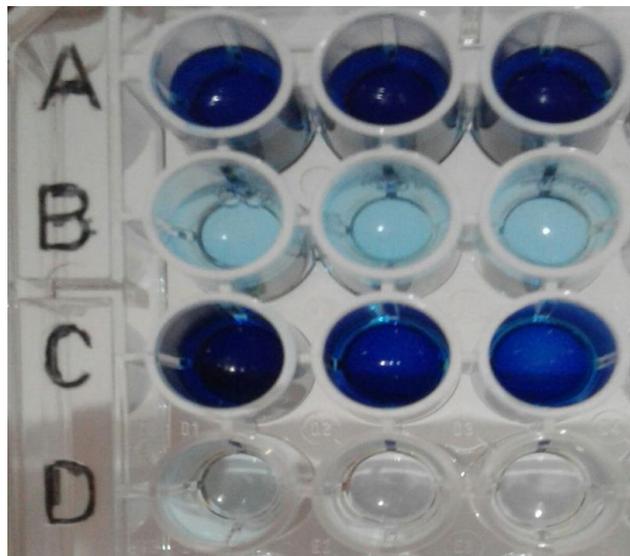


Figure 11. Crystal violet assay of A) MRSA untreated biofilms; B) MRSA biofilms treated with bare VCM C) MRSA biofilms treated with GA D) MRSA biofilms treated with VCM-GAPAH-LPHNPs.

4. Conclusions

There has been an increasing interest in the development of novel nanoantibiotic drug delivery systems due to the global health threat posed by antimicrobial resistance. In this study, the potential to prepare novel pH-responsive lipid-polymer hybrid nanoparticles with co-loaded GA and VCM (VCM-GAPAH-LPHNPs) was explored. A stable VCM-GAPAH-LPHNPs formulation was successfully prepared with favourable particle size, PDI, ZP, morphology, EE% and DL%. The stability of the nanosystems were demonstrated via the analyses of *in silico* studies. *In vitro* biocompatibility and haemolytic studies confirmed the biosafety of VCM-GAPAH-LPHNPs. The release of VCM from VCM-GAPAH-LPHNPs was higher at pH 6 compared to physiological pH 7.4. The *in vitro* antibacterial studies of the VCM-GAPAH-LPHNPs against MRSA revealed an enhanced antibacterial effect at acidic medium compared to pH 7.4, with 75% bacterial elimination in <12 h. The crystal violet assay further validated the antibacterial potential of VCM-GAPAH-LPHNPs by displaying a significantly higher percentage of biofilm eradication when compared to bare VCM, GA and untreated biofilms. Our findings suggest that novel VCM-GAPAH-LPHNPS can potentially be effective in drug delivery to the target site of bacterial infection with low pH and effectively prevent the progression of antimicrobial resistance. In addition, the proposed nanosystem can be used to co-deliver GA with other antimicrobial agents to further explore the antimicrobial potential against other species of bacteria and microorganisms.

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Chapter Four

General conclusions and future recommendations

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Chapter Four – General conclusions and future recommendations

4.1 Introduction

Globally, infectious diseases are one of the top ten leading causes of death. The discovery of antibiotics greatly improved the prevention and control of infections, however, the disadvantages and overuse of conventional antibiotics led to the development of therapy complications and the antimicrobial resistance crisis. One of the major health risks is MRSA infection, which has increased mortality and morbidity rates worldwide. VCM is the last effective resort against MRSA, with the spread of resistance being a serious concern, urging the need to develop strategies to restore VCM efficiency. Although novel nano drug delivery systems have shown enhanced VCM therapy, there is still a need to optimize nano antibiotic carriers to protect and localize VCM during systemic circulation, target VCM release and improve its bioavailability at the infection site. Therefore, current research advances focus on developing pH-responsive nanosystems to improve VCM targeted delivery to the acidic infection sites. LPHNPs are an efficient nanosystem for antibiotic delivery and enhancing antibacterial activity. As per the study aim, the results of the objectives outlined the potential of novel pH-responsive lipid-polymer hybrid nanoparticles co-delivery with VCM and GA to enhance its antibacterial efficacy.

The results pertaining to the study aim and objectives were as follows:

1. To formulate and optimize novel pH-responsive GAPAH lipid-polymer hybrid nanoparticles encapsulating VCM and GA.
2. To optimize and characterize VCM-GAPAH-LPHNPs in terms of particle size, PDI, ZP, pH-responsiveness, morphology, entrapment efficiency and *in vitro* drug release.
3. To assess *in vitro* antibacterial activity of VCM-GAPAH-LPHNPs against MRSA.
4. To identify time-killing kinetics of VCM-GAPAH-LPHNPs against MRSA.
5. To evaluate antibiofilm activity of the novel VCM-GAPAH-LPHNPs.

The main conclusions generated from the research data are summarized below:

- VCM-loaded GA and polyallylamine hydrochloride lipid-polymer hybrid nanoparticles (VCM-GAPAH-LPHNPs) were successfully synthesized.

- The cytotoxicity studies performed by MTT assay on mammalian cell lines (CACO-2 and HEPG2) revealed that VCM-GAPAH-LPHNPs was a biologically safe formulation.
- Spherically shaped VCM-GAPAH-LPHNPs were successfully prepared using micro-emulsion technique. The optimal formulation showed pH-responsiveness in terms of size, PDI and ZP. VCM-GAPAH-LPHNPs size increased from 198.4 ± 0.302 nm at pH 7.4 to 374.8 ± 5.435 nm at pH 6 and 450.4 ± 1.33 nm at pH 4.5 respectively. PDI increased from 0.255 ± 0.003 at pH 7.4 to 0.279 ± 0.054 at pH 6 and 0.377 ± 0.012 at pH 4.5, respectively. Also, ZP switched from -3.8 ± 0.335 mV at pH 7.4 to $+0.44 \pm 0.325$ mV at pH 6 and $+0.693 \pm 0.44$ mV at pH 4.5, respectively. The EE % was found to be 69.46 ± 2.52 .
- *In vitro* drug release studies showed a controlled and pH-dependent VCM release over a period of 48 hours. VCM-GAPAH-LPHNPs had faster drug release at pH 6 compared to pH 7.4.
- *In vitro* antibacterial activity against SA and MRSA confirmed the superiority of VCM-GAPAH-LPHNPs over bare VCM, GA and blank-LPHNPs as VCM-GAPAH-LPHNPs had enhanced and prolonged activity against MRSA at pH 6 and pH 7.4. Moreover, VCM-GAPAH-LPHNPs activity was eight times better against MRSA at pH 6 than at pH 7.4.
- The time-killing assay study showed that VCM-GAPAH-LPHNPs could effectively treat MRSA infections by showing 75% bacterial cell death in less than 12 hours.
- VCM-GAPAH-LPHNPs showed a decrease in crystal violet intensity and were able to eradicate 69 % of MRSA biofilms.
- The findings of this study, therefore, confirmed the potential of the novel GA and polyallylamine hydrochloride for preparation of pH-responsive LPHNPs for enhancing VCM efficacy. In addition, these findings can serve as a basis for future scientific research on the synthesis of novel natural compounds to develop pH-responsive nanocarriers for targeted drug delivery.

4.2 Significance of the findings in the study

The pH-responsive VCM-GAPAH-LPHNPs were designed to improve VCM targeted delivery to acidic target sites of infection for enhancing antibacterial activity. The significance of the findings of the study are as follows:

New Pharmaceutical Products

Novel pH-responsive VCM-GAPAH-LPHNPs formulation were successfully developed in this study, which can stimulate the pharmaceutical industry to develop new pH-responsive materials and medicines to improve antibiotic delivery.

Improved patient therapy and disease treatment

The *in vitro* studies of the developed pH-responsive nanosystem showed enhanced antibiotic activity and sustained VCM release in acidic medium. This nanosystem has the potential to improve bacterial infection therapy by protecting the antibiotic during systemic circulation, target the delivery of optimal antibiotic concentration to infection sites, decrease healthy sites exposure to the antibiotic and enhance bacterial antibiotic uptake. These benefits can result in reducing antibiotic dosing frequency, adverse drug reactions and toxicity. This can lead to improving patient compliance, enhancing antibacterial therapy and combating antimicrobial resistance threats.

Creation of new scientific knowledge

This study can identify new scientific advancements in the preparation and characterization of combinational delivery of antibiotics and natural plant compounds of pH-responsive lipid-polymer hybrid nanoparticles. This can serve as a basis for smart nano delivery systems development and enhance their potential pharmaceutical applications.

Stimulation of new research

The findings can provide potential research advancements to explore the pH-responsive natural compounds i.e., GA, for potential applications in pH-responsive LPHNPs for the co-delivery of antibiotics and natural compounds nanosystems formulation for various drug classes, such as anticancer, antiviral and anti-inflammatory drugs.

4.3 Recommendations for future studies

The present study concluded that the synthesis of novel pH-responsive GA and polyallylamine hydrochloride formulation for VCM and GA targeted co-delivery can enhance treatment of bacterial infections. The following studies are recommended to improve drug targeted delivery via pH-responsive VCM-LPHNPs:

- Other natural compounds, lipids and polymers can be investigated to analyze the effect of lipid-polymer type and ratios on pH-responsiveness, drug entrapment, drug release and antibacterial activity.
- Additional *in vitro* and *in vivo* antimicrobial activity screening against other Gram-positive and Gram-negative bacteria should be done to further assess VCM-GAPAH-LPHNPs spectrum of activity.
- Further *in silico* modeling and simulation studies are required to better understand molecular interactions of VCM-GAPAH-LPHNPs against SA and MRSA bacteria.
- *In vivo* pharmacokinetic profiling could be conducted to provide more information regarding pH-responsive targeted drug delivery, bioavailability and bio-distribution profiles.
- A large-scale production method could be established to influence the development and optimization of the nano antibiotic formulation by local pharmaceutical industries.

4.4 Conclusion

The findings of this study confirmed the potential of the synthesized novel pH-responsive nanoformulation in targeting antibiotic delivery to improve bacterial infection therapy. This study has contributed to the advancement of drug delivery strategies to address the limitations of conventional antibiotic dosage forms and antimicrobial resistance.

Appendix

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Appendix A: Submitted manuscript to the International Journal of Pharmaceutics (Impact factor: 4.845). Reference number: IJP-S-20-04690

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Manuscript Draft

Manuscript Number:

Title: Formulation of pH-responsive lipid-polymer hybrid nanoparticles for co-delivery and enhancement of the antibacterial activity of vancomycin and 18 β -glycyrrhetic acid

Article Type: Research Paper

Section/Category: Pharmaceutical Nanotechnology

Keywords: pH-responsive; lipid-polymer hybrid nanoparticles; vancomycin; methicillin-resistant Staphylococcus aureus; 18 β - glycyrrhetic acid

Corresponding Author: Professor Thirumala Govender, Ph.D.

Corresponding Author's Institution: University of KwaZulu-Natal

First Author: Yajna Jaglal

Order of Authors: Yajna Jaglal; Nawras Osman; Calvin A Omolo; Chunderika Mocktar; Nikita Devnarain; Thirumala Govender, Ph.D.

Abstract: Despite advancement in the control and therapeutics of infectious diseases, antimicrobial resistance remains a global burden. Hence there is a need for novel strategies that improve and potentiate available antibiotics to prevent a regress to a pre-antibiotic era. This study aimed to co-deliver vancomycin (VCM) and 18 β -glycyrrhetic acid (GA) via pH-responsive lipid-polymer hybrid nanoparticles (VCM-GAPAH-LPHNPs) to explore its potential for enhanced activity and targeted delivery of VCM. The stability of VCM-GAPAH-LPHNPs were supported by in silico studies. Biosafe VCM-GAPAH-LPHNPs were prepared using the microemulsion technique. VCM-GAPAH-LPHNPs demonstrated size, polydispersity index, zeta potential and encapsulation efficiency of 198.4 ± 0.302 nm, 0.255 ± 0.003 , -3.8 ± 0.335 mV and 69.46 ± 2.52 %, respectively. In vitro drug release studies revealed that VCM-GAPAH-LPHNPs had sustained and faster release at acidic conditions compared to bare VCM. VCM-GAPAH-LPHNPs also demonstrated greater in vitro antibacterial potential against methicillin-resistant Staphylococcus aureus by 16-fold, when compared to the bare drug. Additionally, the time-killing assay indicated the ability of VCM-GAPAH-LPHNPs to eliminate 75 % of MRSA in less than 12 h. Furthermore, crystal violet assay confirmed VCM-GAPAH-LPHNPs potential to eliminate biofilms. Therefore, these novel LPHNPs may serve as promising nanocarriers for enhancing antibiotic drug delivery and antibacterial activity.

Suggested Reviewers: Yashwanth Pathak
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